

Quiz & Worksheet - Colligative Properties and Raoult's Law

1. What happens to vapor pressure when you add a solute to a solution?

- It lowers the vapor pressure.
- It has no effect.
- It raises the vapor pressure.
- It causes the reaction to reach equilibrium.

2. What is osmotic pressure?

- It is the minimum pressure that must be applied to a solution to stop osmosis from happening.
- It is the maximum pressure that must be applied to a solution to stop osmosis from happening.
- It is the maximum pressure of the vapor over a liquid at equilibrium.
- It is the minimum pressure of the vapor over a liquid at equilibrium.

3.

9.0g of urea (a nonelectrolyte) is dissolved in 360 g water to create a solution. The temperature of the solution is 25.0°C. If the vapor pressure of water at 25.0°C is 23.80 mmHg, calculate the vapor pressure of the urea solution at 25.0°C in mmHg.

The molar mass of urea 60g/mol and water is 18g/mol.

- 0.59
- 23.91
- 23.61
- 3.57

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Colligative Properties Of Solutions Worksheet Answers

**Nivaldo J. Tro, Kathy J. Thrush-
Shaginaw, Mary Beth Kramer**



Colligative Properties Of Solutions Worksheet Answers:

Whispering the Techniques of Language: An Psychological Quest through **Colligative Properties Of Solutions Worksheet Answers**

In a digitally-driven world wherever screens reign great and immediate interaction drowns out the subtleties of language, the profound strategies and mental nuances concealed within words often get unheard. Yet, set within the pages of **Colligative Properties Of Solutions Worksheet Answers** a captivating literary prize pulsing with fresh emotions, lies a fantastic journey waiting to be undertaken. Written by a talented wordsmith, this wonderful opus encourages visitors on an introspective journey, gently unraveling the veiled truths and profound affect resonating within ab muscles material of each word. Within the emotional depths with this moving review, we will embark upon a heartfelt exploration of the book is core styles, dissect its captivating publishing type, and yield to the powerful resonance it evokes heavy within the recesses of readers hearts.

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