
IGCSE CHEMISTRY NOTES

Syllabus

[IGCSE Chemistry \(0620\) Syllabus](#)

The particulate nature of matter

- Distinguishing properties

SOLID	LIQUID	GAS
<ul style="list-style-type: none">• Strong forces of attraction between particles• Fixed pattern (lattice)• Atoms vibrate but can't change position \therefore fixed volume and shape	<ul style="list-style-type: none">• Weaker attractive forces than solids• No fixed pattern, liquids take up the shape of their container• Particles slide past each other.	<ul style="list-style-type: none">• Almost no intermolecular forces• Particles far apart, and move quickly• Collide with each other and bounce in all directions

- Structure
 - Solids - Particles are arranged in a fixed particle or lattice. They cannot leave their positions. They have a to and fro motion called vibration
 - Liquid - The particles in a liquid can move around and slide past each other. They are close but have no lattice. 3R motion - Random, Rapid, Rectilinear
 - Gases - There are almost no forces holding them together. 3R motion - Random, Rapid, Rectilinear

Cambridge Chemistry Notes

Judd E. Hollander



Cambridge Chemistry Notes:

Cambridge Chemistry Notes Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Cambridge Chemistry Notes**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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