

## MT 18 STANDARD WATERS

## INTRODUCTION (Note 1)

Instructions for the preparation of Standard Waters and of waters of any desired hardness are given in CIPAC Monograph 1. For the Standard Waters A to G, stock solutions of  $\text{Ca}^{++}$  and  $\text{Mg}^{++}$  are prepared; the working solutions are then made by dilution of these stock solutions.

All solutions should be stored in polyethylene containers.

Working solutions should be freshly prepared immediately before use.

A method for the determination of the hardness of water by EDTA titration is given in MT 73.

## 18.1 Preparation of Standard Waters A to G (MT 18.1.1 to 18.1.7)

## REAGENTS

*Calcium carbonate* ( $\text{CaCO}_3$ ) containing not less than 99% calcium carbonate; dry for 2 h at 105 °C before use

*Magnesium oxide* ( $\text{MgO}$ ) containing not less than 99% magnesium oxide; dry for 2 h at 105 °C before use

*Sodium hydrogen carbonate* ( $\text{NaHCO}_3$ ) containing not less than 99% sodium hydrogen carbonate

*Ammonia solution* ( $\text{NH}_4\text{OH}$ ) approximately 1 mol/l (1N)

*Hydrochloric acid c* ( $\text{HCl}$ ) : 1 mol/l (1N) and 0.1 mol/l (0.1N) standardized solutions; RE 14.2 and 14.1

*Sodium hydroxide c* ( $\text{NaOH}$ ) : 0.1 mol/l (0.1N) standardized solution; RE 25.1

*Methyl red indicator* 0.1% m/v aqueous solution; RE 18

## STOCK SOLUTIONS

*Solution 1. 0.04M  $\text{Ca}^{++}$  solution*

Weigh accurately calcium carbonate (4.000 g), and transfer to a 500 ml conical flask with a minimum of distilled water. Place a small filter funnel in the mouth of the flask and add slowly 1 mol/l hydrochloric acid (82.0 ml, measured from a burette), swirling the contents. When all the calcium carbonate has dissolved, dilute the solution to approximately 400 ml with distilled water, and boil gently to expel excess carbon dioxide. Cool the solution, add methyl red (2 drops) and neutralize to an intermediate orange colour with 1 mol/l ammonia solution, added drop wise.

# Cipac Standard Water Pdfslibforme

**Mark Steyvers**



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