

## Quiz & Worksheet - Colligative Properties and Raoult's Law

1. What happens to vapor pressure when you add a solute to a solution?

- It lowers the vapor pressure.
- It has no effect.
- It raises the vapor pressure.
- It causes the reaction to reach equilibrium.

2. What is osmotic pressure?

- It is the minimum pressure that must be applied to a solution to stop osmosis from happening.
- It is the maximum pressure that must be applied to a solution to stop osmosis from happening.
- It is the maximum pressure of the vapor over a liquid at equilibrium.
- It is the minimum pressure of the vapor over a liquid at equilibrium.

3.

9.0g of urea (a nonelectrolyte) is dissolved in 360 g water to create a solution. The temperature of the solution is 25.0°C. If the vapor pressure of water at 25.0°C is 23.80 mmHg, calculate the vapor pressure of the urea solution at 25.0°C in mmHg.

The molar mass of urea 60g/mol and water is 18g/mol.

- 0.59
- 23.91
- 23.61
- 3.57

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# Colligative Properties Of Solutions Worksheet Answers

**Yijin Wang**



**Colligative Properties Of Solutions Worksheet Answers:**

## Reviewing **Colligative Properties Of Solutions Worksheet Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Colligative Properties Of Solutions Worksheet Answers**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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