

Chemistry Chapter 5 Exam

Multiple Choice

Identify the letter of the choice that best completes the statement or answers the question.

- _____ 1. The idea of arranging the elements in the periodic table according to their chemical and physical properties is attributed to
a. Mendeleev. c. Bohr.
b. Moseley. d. Ramsay.
- _____ 2. Mendeleev left spaces in his periodic table and predicted several elements and their
a. atomic numbers. c. properties.
b. colors. d. radioactivity.
- _____ 3. Mendeleev attempted to organize the chemical elements based on their
a. symbols. c. atomic numbers.
b. properties. d. electron configurations.
- _____ 4. Mendeleev noticed that properties of elements usually repeated at regular intervals when the elements were arranged in order of increasing
a. atomic number. c. reactivity.
b. density. d. atomic mass.
- _____ 5. Mendeleev is credited with developing the first successful
a. periodic table.
b. method for determining atomic number.
c. test for radioactivity.
d. use of X rays.
- _____ 6. Mendeleev did not always list elements in his periodic table in order of increasing atomic mass because he grouped together elements with similar
a. properties. c. densities.
b. atomic numbers. d. colors.
- _____ 7. In developing his periodic table, Mendeleev listed on cards each element's name, atomic mass, and
a. atomic number. c. isotopes.
b. electron configuration. d. properties.
- _____ 8. Mendeleev predicted that the spaces in his periodic table represented
a. isotopes. c. permanent gaps.
b. radioactive elements. d. undiscovered elements.
- _____ 9. Mendeleev's table was called periodic because the properties of the elements
a. showed no pattern.
b. occurred at repeated intervals called periods.
c. occurred at regular time intervals called periods.
d. were identical.
- _____ 10. The person whose work led to a periodic table based on increasing atomic number was
a. Moseley. c. Rutherford.
b. Mendeleev. d. Cannizzaro.
- _____ 11. Moseley's work led to the realization that elements with similar properties occurred at regular intervals when the elements were arranged in order of increasing
a. atomic mass. c. radioactivity.
b. density. d. atomic number.
- _____ 12. Who used his experimental evidence to determine the order of the elements according to atomic number?
a. Meyer c. Stas

Modern Chemistry Chapter 5 Section 1 Answers

Didier Musso



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