

TEACHING TRANSPARENCY WORKSHEET

9

Types of Matter

**Use with Chapter 3,
Section 3.4**

1. Into what two broad classes can all matter be divided?

2. What is the difference between a mixture and a pure substance?

3. What is the difference between a compound and an element?

4. Can a compound be a heterogeneous mixture? Explain your answer by referring to the diagram.

5. A list of compounds and elements is given below. Circle the substances that are elements.

gold	aluminum	water
sugar	oxygen	platinum
salt	chlorine	brass

6. How can you tell the difference between a homogeneous mixture and a heterogeneous mixture?

7. Label each mixture below as either homogeneous or heterogeneous.

<ol style="list-style-type: none"> a. air _____ b. clay _____ c. homemade lemonade (with pulp) _____ d. oatmeal raisin cookie _____ 	<ol style="list-style-type: none"> e. finger paint _____ f. vinegar _____ g. soil _____
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8. List three methods that are commonly used to separate mixtures into their component substances.

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Teaching Transparency Chemistry Chapter 19

M Carnoy



Teaching Transparency Chemistry Chapter 19:

Reviewing **Teaching Transparency Chemistry Chapter 19**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Teaching Transparency Chemistry Chapter 19**," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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