

Chapter Test B

Chapter: Chemical Equations and Reactions

Match a name space (possibly writing the name of the react or product molecule) completed with reactants/products atoms/molecules.

- The product(s) of a chemical reaction will be represented by a double arrow pointing towards the right in the equation of a reaction.
 - reactants
 - products
 - solvents
- An ionic chemical equation is represented by adding the
 - charges
 - structure of the particles
 - number of particles
- An ionic equation shows a reaction by adding particles that are in the
 - state
 - aqueous or ionic
 - solid state
- The activity for precipitation reactions are used to predict whether
 - soluble ionic compounds will react to form an insoluble precipitate
 - insoluble ionic compounds will react to form a soluble precipitate
 - soluble ionic compounds will react to form a precipitate
 - insoluble ionic compounds will react to form a precipitate
- The coefficients are chemical equations
 - indicate the number of moles of each substance that react
 - show the number of grams of each substance that react
 - show the number of moles of the substances
 - show the relative proportions of each atom
- If you're to balance the chemical equation with one variable, then
 - X will replace one of the coefficients
 - X will replace one of the subscripts
 - X will replace one of the elements
 - X will replace both of the atoms

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Glencoe Law and Ethics Medical Careers Karen Judson, Sharon Blesie-Hicks, 1998-12

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Holt Science and Technology Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001-07

American Physical Education Review, 1924 Includes abstracts of magazine articles and Book reviews

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