

**TEACHING TRANSPARENCY WORKSHEET**

# Types of Matter

**Use with Chapter 3,  
Section 3.4**

1. Into what two broad classes can all matter be divided?

\_\_\_\_\_

2. What is the difference between a mixture and a pure substance?

\_\_\_\_\_

3. What is the difference between a compound and an element?

\_\_\_\_\_

4. Can a compound be a heterogeneous mixture? Explain your answer by referring to the diagram.

\_\_\_\_\_

\_\_\_\_\_

5. A list of compounds and elements is given below. Circle the substances that are elements.

- |       |          |          |
|-------|----------|----------|
| gold  | aluminum | water    |
| sugar | oxygen   | platinum |
| salt  | chlorine | brass    |

6. How can you tell the difference between a homogeneous mixture and a heterogeneous mixture?

\_\_\_\_\_

\_\_\_\_\_

7. Label each mixture below as either homogeneous or heterogeneous.

- |  |                       |
|--|-----------------------|
| a. air _____                           | e. finger paint _____ |
| b. clay _____                          | f. vinegar _____      |
| c. homemade lemonade (with pulp) _____ | g. soil _____         |
| d. oatmeal raisin cookie _____         |                       |

8. List three methods that are commonly used to separate mixtures into their component substances.

\_\_\_\_\_

\_\_\_\_\_

# Teaching Transparency Chemistry Chapter 19

**K. V. Sane**



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