



COVALENT BONDING

Practice Problems

Do your answers to the following problems.

SECTION 5.1 MOLECULAR COMPOUNDS

- Classify each of the following as either an ionic or a molecular compound.
 - NaCl
 - H_2O
 - HCl
 - KNO_3
 - C_2H_6
 - H_2SO_4
- Write the formula for the following compounds:
 - nitrogen dioxide
 - carbon dioxide
 - carbon monoxide
 - hydrogen peroxide
 - hydrogen cyanide
 - hydrogen sulfide

- Write the molecular formula for each of the following compounds:
 - nitrogen dioxide
 - carbon dioxide
 - carbon monoxide
 - hydrogen peroxide
 - hydrogen cyanide
 - hydrogen sulfide
- Write the molecular formula for each of the following compounds:
 - nitrogen dioxide
 - carbon dioxide
 - carbon monoxide
 - hydrogen peroxide
 - hydrogen cyanide
 - hydrogen sulfide

SECTION 5.2 THE NATURE OF COVALENT BONDING

- Draw the Lewis structure for hydrogen fluoride. $\text{H}-\text{F}$
- Draw the Lewis structure for carbon dioxide. $\text{O}=\text{C}=\text{O}$
- Draw the Lewis structure for methane. C with four H atoms bonded to it.
- Draw the Lewis structure for acetylene. $\text{H}-\text{C}\equiv\text{C}-\text{H}$
- Draw the Lewis structure for hydrogen cyanide. $\text{H}-\text{C}\equiv\text{N}$



SECTION 5.3 BONDING: THE ORBITALS

- Which orbital and suborbital are represented by each of the following?
 - $2s$
 - $2p$
 - $3s$
 - $3p$
 - $4s$
 - $4p$
- Which orbital and suborbital are represented by each of the following?
 - $2s$
 - $2p$
 - $3s$
 - $3p$
 - $4s$
 - $4p$

Section 8 Covalent Bonding Answers

Seager/Slabaugh



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Solutions Manual to Accompany General Chemistry, Fourth Edition, and General Chemistry with Qualitative Analysis, Fourth Edition, Whitten, Gailey, Davis Yi-Noo Tang, Wendy L. Keeney-Kennicutt, 1992 **Chemistry** John W. Moore, Conrad L. Stanitski, Peter C. Jurs, 2002 CHEMISTRY THE MOLECULAR SCIENCE is intended to help students develop a broad overview of chemistry and chemical reactions an understanding of the most important concepts and models that chemists and those in chemistry related fields use an appreciation of the many ways chemistry impacts our daily lives the ability to apply the facts concepts and models of chemistry appropriately to new situations in chemistry other sciences and engineering and to other disciplines *Student Solutions Manual for Whitten, Davis, Peck, and Stanley's General Chemistry, 7th Ed* Wendy L. Keeney-Kennicutt, 2004 Contains answers and solutions to all even numbered end of chapter exercises Solutions are divided by section for easy reference by students **Study Guide and Solutions Manual for Organic Chemistry Digital Update** K. Peter C. Vollhardt, Neil E. Schore, 2024-03-25 Written by Organic Chemistry coauthor Neil Schore this invaluable manual includes chapter introductions that highlight new materials chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today Seager/Slabaugh, 2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application Brown's Introduction to Organic Chemistry, Global Edition William H. Brown, Thomas Poon, 2017-06-06 Brown s Introduction to Organic Chemistry provides an introduction to organic chemistry for students who require the fundamentals of organic chemistry as a requirement for their major It is most suited for a one semester organic chemistry course In an attempt to highlight the relevance of the material to students the authors place a strong emphasis on showing the interrelationship between organic chemistry and other areas of science particularly the biological and health sciences The text illustrates the use of organic chemistry as a tool in these sciences it also stresses the organic compounds both natural and synthetic that surround us in everyday life in pharmaceuticals plastics fibers agrochemicals surface coatings toiletry preparations and cosmetics food additives adhesives and elastomers **Student Solutions Manual, Chemistry, Principles & Practice, Second Edition** John M. DeKorte, 1997 **Chapter Resource 2 Chemistry of Life Biology** Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004 *Oswaal CBSE Sample Question Papers Class 10 Science Book (For 2024 Exam)* Oswaal Editorial Board, 2023-08-09 *DAT 2017-2018 Strategies, Practice & Review with 2 Practice Tests* Kaplan Test Prep, 2016-10-04 Kaplan s DAT 2017 2018 Strategies Practice Review features the realistic practice test taking strategies and expert guidance

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