

CHEMISTRY LABORATORY REPORT
SK025



NAME : Yuelina alp Muliawati
NO. OF EXPERIMENT : 4
DATE OF EXPERIMENT : 25/4/2022
PRACTICUM : HVP2
LECTURER'S NAME : Mdm. Gan Fie Chuen

TITLE	Reactions of Aliphatic and Aromatic Hydrocarbons
OBJECTIVE	<ol style="list-style-type: none">1. To study the chemical properties of an alkane, alkene and arene2. To differentiate an alkane from alkene and arene
THEORY	<p>Hydrocarbons are organic compounds that contain only carbon and hydrogen. Alkanes which are also known as paraffins are saturated hydrocarbons. They do not contain double or triple bonds. Hence, alkanes are relatively inert to chemical reactions. Alkanes undergo free radical substitution reaction.</p> <p>Alkenes are unsaturated hydrocarbons, with at least one double bond between two carbon atoms. Alkenes can easily undergo addition reactions at the C=C bond. For example, alkenes undergo hydrogenation and halogenation to form alkanes and dihalides respectively. Alkenes can also react with potassium permanganate solution in two different conditions.</p> <p>Arenes are aromatic hydrocarbons with stable molecular structures. Although arenes have a very high degree of unsaturation, they are relatively inert towards all addition reactions except at very high pressure and temperature. Arenes undergo electrophilic aromatic substitution reactions in the presence of a Lewis acid catalyst.</p>

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Benjamin Alexander



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