

# QUALITATIVE ANALYSIS (CATIONS)

Group	Group reagent	Cations (Basic radicals)	Composition & colour of precipitate
1	dil HCl	$Pb^{2+}$ (w)	$PbCl_2$ ; white
		$Ag^+$	$AgCl$ ; white (adv)
		$Hg_2^{2+}$	$Hg_2Cl_2$ ; white
2(A)	$H_2S$ in presence of dil HCl (insoluble in YAS)	$Pb^{2+}$ (w)	$PbS$ ; Black
		$Cu^{2+}$ (w)	$CuS$ ; Black
		$Hg^{2+}$	$HgS$ ; Black (adv)
		$Bi^{3+}$	$Bi_2S_3$ ; Black
		$Co^{2+}$ $Cd^{2+}$	$CoS$ ; yellow
2(B)	$H_2S$ in presence of dil HCl (soluble in YAS)	$As^{3+}, Sb^{3+}$	$As_2S_3$ ; yellow, $Sb_2S_3$ ; orange
		$Sn^{2+}, Sn^{4+}$	$SnS$ ; brown, $SnS_2$ ; yellow
3	$NH_4OH$ in presence of $NH_4Cl$	$Fe^{3+}$ (w)	$Fe(OH)_3$ ; red brown
		$Al^{3+}$ (w)	$Al(OH)_3$ ; gelatinous white
		$Cr^{3+}$	$Cr(OH)_3$ ; green (adv)
4	$H_2S$ in presence of $NH_4OH$ & $NH_4Cl$	$Zn^{2+}$ (w)	$ZnS$ ; white
		$Ni^{2+}$ (w)	$NiS$ ; black
5	$(NH_4)_2CO_3$ in presence of $NH_4OH$ <small>Mns (Bull/Pine)</small>	$Mn^{2+}$ (w)	$MnCO_3$ ; black, $MnS$ ; black
		$Ba^{2+}$ (w)	$BaCO_3$ ; white
		$Ca^{2+}$ (w)	$CaCO_3$ ; white
		$Sr^{2+}$	$SrCO_3$ ; white
6	$Na_2HPO_4$ in presence of $NH_4OH$	$Mg^{2+}$ (w)	$Mg(NH_4)PO_4$ ; white

# Qualitative Analysis Of Cations Lab Report Answers

**George H. Schenk**



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Laboratory Manual for Principles of General Chemistry Jo Allan Beran,2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures

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**Chemistry in the Laboratory** James M. Postma,Julian L. Robert,J. Leland Hollenberg,2004-03-12 This clearly written class tested manual has long given students hands on experience covering all the essential topics in general chemistry Stand alone experiments provide all the background introduction necessary to work with any general chemistry text This revised edition offers new experiments and expanded information on applications to real world situations

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**Principles Of Qualitative Inorganic Analysis** Huda Alhasan; Nadiyah,Nadiyah Alahmadi,Huda S Alhasan,2021-01-26 This book provides notes for basic laboratory experiments in qualitative analysis of cations The book introduces readers to basic methods and laboratory safety Subsequent chapters cover six groups of cations Each chapter explains important details that are required to understand how a particular analytical method works for detecting cations in samples starting from sedimentation and ending with the identification Key Features Simple reader friendly format introductory notes and summary Covers several groups of metals Appendix for handy reference with tables and references This is a useful textbook for early chemistry students and teachers as it equips the readers with sufficient information required to analyze chemical samples and deduce the presence of specific cations as part of laboratory coursework

**Nonsulfide Qualitative Analysis of Cations; Separating & Identifying Representative Cations from Group A-E** Judith C. Foster,1995-01-01 Qualitative Analysis as a Laboratory Basis for the Study of General Inorganic Chemistry William Conger Morgan,1906

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