

Bachelor Level/B.Pharm/First Year/TU/IOM Time: 3 hours  
 Pharmaceutical Chemistry-403A Th. (New Course) Full Marks: 80

Candidates are required to give their answers in their own words as far as practicable.

The figures in the margin indicate full marks.  
 Attempt all the questions.

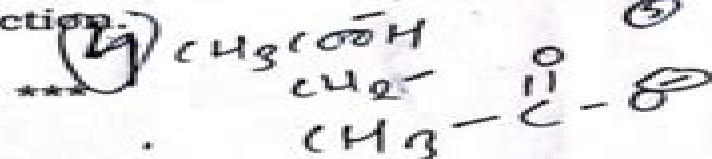
1. What is the importance of limit test and assay determination in pharmaceuticals? How is the limit test of sulphate performed? (5) 4+6=10
2. What is the basic objective of replacement therapy? Explain the method of preparation, test of purity, uses and assay of sodium lactate. (5) 3+7=10
3. Explain the method for conducting assay of: Na<sub>2</sub> 2x5=10
  - a. Potassium permanganate
  - b. Ferric ammonium citrate
4. Define topical agents. Write the applications of protectives, astringents and antimicrobial agents with two suitable examples of each. (5) 2+8=10
5. Explain why aqueous base brings substitution reaction and alcoholic base brings elimination reaction. Explain the mechanism of E1, E2 and E1CB reaction. (5) 4+6=10
6. Give the mechanism for: (5) 2x5=10
  - a. Reaction of ethyl acetate with the mixture of ethanol and sodium ethoxide
  - b. Reaction of two molecules of acetaldehyde in presence of sodium hydroxide
7. Define the following with suitable examples: (5) 4+3+3=10
  - a. Conformational isomers
  - b. Meso compounds
  - c. Optical activity
8. What is meant by electrophilic substitution reaction? Explain the effect of electron releasing and withdrawing groups in electrophilic substitution reaction. Give mechanism for any one of the electrophilic substitution reaction. (5) 2+4+4=10

CO<sub>2</sub>Na

C<sub>6</sub>H<sub>5</sub>

NH<sub>2</sub><sup>+</sup> NH  
 NH<sub>2</sub><sup>-</sup>

H-  
 Zn-S



# 3rd Year B Pharmacy Question Paper

**R Sandford**



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## **Table of Contents 3rd Year B Pharmacy Question Paper**

1. Understanding the eBook 3rd Year B Pharmacy Question Paper
  - The Rise of Digital Reading 3rd Year B Pharmacy Question Paper
  - Advantages of eBooks Over Traditional Books
2. Identifying 3rd Year B Pharmacy Question Paper
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an 3rd Year B Pharmacy Question Paper
  - User-Friendly Interface
4. Exploring eBook Recommendations from 3rd Year B Pharmacy Question Paper
  - Personalized Recommendations
  - 3rd Year B Pharmacy Question Paper User Reviews and Ratings
  - 3rd Year B Pharmacy Question Paper and Bestseller Lists

5. Accessing 3rd Year B Pharmacy Question Paper Free and Paid eBooks
  - 3rd Year B Pharmacy Question Paper Public Domain eBooks
  - 3rd Year B Pharmacy Question Paper eBook Subscription Services
  - 3rd Year B Pharmacy Question Paper Budget-Friendly Options
6. Navigating 3rd Year B Pharmacy Question Paper eBook Formats
  - ePub, PDF, MOBI, and More
  - 3rd Year B Pharmacy Question Paper Compatibility with Devices
  - 3rd Year B Pharmacy Question Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of 3rd Year B Pharmacy Question Paper
  - Highlighting and Note-Taking 3rd Year B Pharmacy Question Paper
  - Interactive Elements 3rd Year B Pharmacy Question Paper
8. Staying Engaged with 3rd Year B Pharmacy Question Paper
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers 3rd Year B Pharmacy Question Paper
9. Balancing eBooks and Physical Books 3rd Year B Pharmacy Question Paper
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection 3rd Year B Pharmacy Question Paper
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine 3rd Year B Pharmacy Question Paper
  - Setting Reading Goals 3rd Year B Pharmacy Question Paper
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 3rd Year B Pharmacy Question Paper
  - Fact-Checking eBook Content of 3rd Year B Pharmacy Question Paper
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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