

Quality Control Interview Questions & Answers

Redox titrations: Where the titrant is oxidizing agents or reducing agents.

Q47. What is the Ultraviolet (UV) and visible spectroscopy range?

The range of UV Spectroscopy is 200-400 nm, and visible spectroscopy ranges from 400- 800 nm.

Q48. What is the use of UV Spectroscopy?

Spectroscopy can be used to detect functional groups, and impurities, and perform qualitative and quantitative analyses.

Q49. What is meant by the solution?

A solution is a mixture of liquids, gases, and solids, the solution consists of many different types of solutes like salts, oxygen, and organic molecules.

Q50. Describe the saturated and unsaturated solutions

A saturated solution is defined as a solution in which a solvent is not capable of dissolving any more solute at a given temperature.

At a given temperature, an unsaturated solution is one in which the solvent is Capable of dissolving any extra solute.

Q51. What is the difference between qualitative and quantitative analysis?

Qualitative analysis involves the identification of the compound or chemical based on their chemical (absorption, emission) or physical properties, e.g., melting point and boiling point.

Quantitative analysis: this involves estimation or determination of the concentration or amount of the chemical compounds or components.

Q52. Explain the principle of ultraviolet spectroscopy?

Ultraviolet spectroscopy uses light in the UV part of the electromagnetic spectrum. UV absorption spectra form when the outer electrons of a molecule or an atom absorb energy and move from a lower to a higher energy level. The wavelength absorbance of each molecule is unique.

Q53. Define Molarity?

A number of moles of solute per litre solution. Molarity is denoted with a capital "M".

Q54. Define molality?

The number of moles of solute per kilogram solvent. It is denoted with a small "m".

Q55. Define normality?

The number of moles equivalent per litre solution.

Q56. What are buffer solutions?

A buffer solution is an aqueous solution consisting of a mixture of a weak acid and its conjugate base or vice versa.

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Intelligent Internal Control and Risk Management Matthew Leitch, 2016-05-23 Many people in organizations resent internal control and risk management these two processes representing unwelcome tasks to be completed for the benefit of auditors and regulators Over the last few years this perception has been heightened by the disastrous implementation of section 404 of the Sarbanes Oxley Act of 2002 which is generally regarded as having been too expensive for the benefits it has brought This important book offers a way of improving this prevailing perception and increasing the value of control and risk management by bringing creativity and design skills to the fore The value of risk and control activities is often limited by the value of the control ideas available and so Matthew Leitch provides an arsenal of 60 high performance control mechanisms These include several alternative ways to design controls and control systems as well as providing controls for monitoring and audit controls for accelerated learning and techniques for finding and recovering cash This design material is combined with insights into the psychology of risk control strategies for encouraging helpful behaviour and enabling change and a surprisingly simple integration of internal control with risk management The book is realistic practical original and easier reading than most in the field The material is not specific to any one country and has international appeal for internal auditors and all those concerned with risk management corporate governance and security

Internal control course, 1984 **Internal Control Course** United States. Department of Defense, 1984 **Resource Management Journal**, 1984

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