

**AP Chemistry Unit 3 Practice Test**  
**Solids, Liquids, Gases and Solutions**

Name \_\_\_\_\_  
Date \_\_\_\_\_ Block \_\_\_\_\_

1. A real gas would act most ideal at  
(A) 1 atm and 273 K (D) 0.5 atm and 546 K  
(B) 10 atm and 547 K (E) 0.5 atm and 273 K  
(C) 10 atm and 273 K
2. Zinc reacts with aqueous sulfuric acid to form hydrogen gas:  
$$\text{Zn (s)} + \text{H}_2\text{SO}_4 \text{ (aq)} \rightarrow \text{ZnSO}_4 \text{ (aq)} + \text{H}_2 \text{ (g)}$$
  
In an experiment, 201 mL of wet  $\text{H}_2$  is collected over water at  $27^\circ\text{C}$  and a barometric pressure of 733 torr. The vapor pressure of water at  $27^\circ\text{C}$  is 26.74 torr. The partial pressure of hydrogen in this experiment is \_\_\_\_\_ atm.  
(A) 0.929 (D) 1.00  
(B) 760 (E) 0.964  
(C) 706
3. A mixture of gases contains 1.5 moles of oxygen, 3.0 moles of nitrogen, and 0.5 mole of water vapor. If the total pressure is 700 mmHg, what is the partial pressure of the nitrogen gas?  
(A) 70 mmHg (D) 350 mmHg  
(B) 210 mmHg (E) 420 mmHg  
(C) 280 mmHg
4. Equal numbers of moles of  $\text{CO}_2(\text{g})$ ,  $\text{N}_2(\text{g})$ , and  $\text{NH}_3(\text{g})$  are placed in a sealed vessel at room temperature. If the vessel has a pinhole-size leak, which of the following will be true after some of the gas mixture has effused?  
I. The mole fraction of  $\text{CO}_2$  in the sample will increase.  
II. The  $\text{N}_2$  will effuse the fastest since it is the lightest.  
III. All gases will effuse at the same rate since the temperature is held constant.  
(A) I only (C) I and II (E) I, II, and III  
(B) III only (D) II and III
5. A sample of oxygen gas was found to effuse at a rate equal to two times that of an unknown gas. The molecular weight of the unknown gas is \_\_\_\_\_ g/mol.  
(A) 128  
(B) 8  
(C) 16  
(D) 64

# Ap Chemistry Practice Test And Answers

**Jessica J Manson**



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## **Table of Contents Ap Chemistry Practice Test And Answers**

1. Understanding the eBook Ap Chemistry Practice Test And Answers
  - The Rise of Digital Reading Ap Chemistry Practice Test And Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Ap Chemistry Practice Test And Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Ap Chemistry Practice Test And Answers
  - User-Friendly Interface

4. Exploring eBook Recommendations from Ap Chemistry Practice Test And Answers
  - Personalized Recommendations
  - Ap Chemistry Practice Test And Answers User Reviews and Ratings
  - Ap Chemistry Practice Test And Answers and Bestseller Lists
5. Accessing Ap Chemistry Practice Test And Answers Free and Paid eBooks
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6. Navigating Ap Chemistry Practice Test And Answers eBook Formats
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  - Ap Chemistry Practice Test And Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Ap Chemistry Practice Test And Answers
  - Highlighting and Note-Taking Ap Chemistry Practice Test And Answers
  - Interactive Elements Ap Chemistry Practice Test And Answers
8. Staying Engaged with Ap Chemistry Practice Test And Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Ap Chemistry Practice Test And Answers
9. Balancing eBooks and Physical Books Ap Chemistry Practice Test And Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Ap Chemistry Practice Test And Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Ap Chemistry Practice Test And Answers
  - Setting Reading Goals Ap Chemistry Practice Test And Answers
  - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Ap Chemistry Practice Test And Answers
  - Fact-Checking eBook Content of Ap Chemistry Practice Test And Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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