

Colorimetric Analysis Tutorial Answers

1. a) The limitations of the Beer-Lambert Law are:

- Deviations in absorption coefficients at high concentrations (>0.01M) due to electrostatic interactions between molecules in close proximity
- Scattering of light due to particulates in the sample
- Fluorescence or phosphorescence of the sample
- Refractive index changes with analyte concentration
- Shifts in chemical equilibrium as a function of concentration
- Non-monochromatic radiation (deviations can be minimized by using a relatively flat part of the absorption spectrum such as the maximum of an absorption band)
- Stray light
- Unmatched cells

See D A Skoog, F J Holler, S R Crouch, (2007) "Principles of Instrumental Analysis", 6th Ed., p 336-343 for a full description.

b) Colorimetric analysis of mixtures requires/assumes no interaction between the various species present i.e. they do not change each other's colour/absorbance spectrum.

2. Break the problem down into steps and use equations for propagation of errors.

	Value	SD
A_1	0.312	0.003
A_2	0.092	0.002
l	1.000	0.005
ϵ_{A1}	70.660	2.165
ϵ_{A2}	3.510	0.197
ϵ_{B1}	16.910	0.786
ϵ_{B2}	107.230	3.055

Step	Value	Error
$\epsilon_{B1} A_2$	1.5557	0.0798
$\epsilon_{B2} A_1$	33.4558	1.0060
$\epsilon_{A2} \epsilon_{B1}$	59.3541	4.3254
$\epsilon_{A1} \epsilon_{B2}$	7576.8718	317.0067
$\epsilon_{B1} A_2 - \epsilon_{B2} A_1$	-31.9000	1.0091
$\epsilon_{A2} \epsilon_{B1} - \epsilon_{A1} \epsilon_{B2}$	7517.5177	317.0362
c_A	0.004243	0.0002247

$$= 1.5557 \times \sqrt{[(0.786/16.91)^2 + (0.002/0.092)^2]}$$

$$= 33.4558 \times \sqrt{[(3.055/107.23)^2 + (0.003/0.312)^2]}$$

$$= 59.3541 \times \sqrt{[(0.197/3.51)^2 + (0.786/16.91)^2]}$$

$$= 7576.8718 \times \sqrt{[(2.165/70.66)^2 + (3.055/107.23)^2]}$$

$$= \sqrt{[(0.0798)^2 + (1.0060)^2]}$$

$$= \sqrt{[(4.3254)^2 + (317.0067)^2]}$$

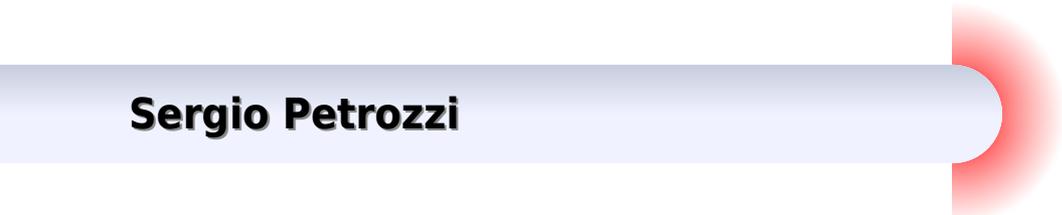
$$= 0.004243 \times \sqrt{[(1.0091/-31.9000)^2 + (317.0362/-7517.5177)^2 + (0.005/1)^2]}$$

Step	Value	Error
$\epsilon_{A1} A_2$	1.0951	0.0624
$\epsilon_{A2} A_1$	6.5007	0.2442
$\epsilon_{B1} \epsilon_{A2}$	59.3541	4.3254
$\epsilon_{B2} \epsilon_{A1}$	7576.8718	317.0067
$\epsilon_{A1} A_2 - \epsilon_{A2} A_1$	-5.4056	0.2521
$\epsilon_{B2} \epsilon_{A1} - \epsilon_{B1} \epsilon_{A2}$	-7517.5177	317.0362
c_B	0.000719	0.000045

Therefore: $c_A = 0.00424 \pm 0.00022$ M and $c_B = 0.00072 \pm 0.00005$ M

Colorimetric Analysis Lab Answers

Sergio Petrozzi



Colorimetric Analysis Lab Answers:

Practical Instrumental Analysis Sergio Petrozzi, 2012-09-24 This practical book in instrumental analytics conveys an overview of important methods of analysis and enables the reader to realistically learn the principally technology independent working techniques the analytical chemist uses to develop methods and conduct validation What is to be conveyed to the student is the fact that analysts in their capacity as problem solvers perform services for certain groups of customers i e the solution to the problem should in any case be processed in such a way as to be fit for purpose The book presents sixteen experiments in analytical chemistry laboratory courses They consist of the classical curriculum used at universities and universities of applied sciences with chromatographic procedures atom spectrometric methods sensors and special methods e g field flow fractionation flow injection analysis and N determination according to Kjeldahl The carefully chosen combination of theoretical description of the methods of analysis and the detailed instructions given are what characterizes this book The instructions to the experiments are so detailed that the measurements can for the most part be taken without the help of additional literature The book is complemented with tips for effective literature and database research on the topics of organization and the practical workflow of experiments in analytical laboratory on the topic of the use of laboratory logs as well as on writing technical reports and grading them Evaluation Guidelines for Laboratory Experiments A small introduction to Quality Management a brief glance at the history of analytical chemistry as well as a detailed appendix on the topic of safety in analytical laboratories and a short introduction to the new system of grading and marking chemicals using the Globally Harmonized System of Classification and Labelling of Chemicals GHS round off this book This book is therefore an indispensable workbook for students internship assistants and lecturers in the area of chemistry biotechnology food technology and environmental technology in the basic training program of analytics at universities and universities of applied sciences

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Pharmacology Thesis Earvin P Eugene,2020-05-13 *Metabolism of Drugs in the Liver of Rats Evaluating Cytochrome P450 Activity* Stony Brook University information Assessment of Sampling Error Associated with Collection and Analysis of Soil Samples at Explosives-Contaminated Sites , Technical Association of the Pulp and Paper Industry ,1970 *Operation of Wastewater Treatment Plants* Kenneth D. Kerri,2008 Clinical Laboratory Medicine Richard C. Tilton,1992 *Leather Industries Laboratory Book of Analytical and Experimental Methods* Henry Richardson Procter,1908

Standard Instrumentation Questions and Answers: Measuring systems Stephen Michael Elonka,Alonzo Ritter Parsons,1962 *Cumulated Index Medicus* ,1980 *Chemical Age* ,1922 Medical Laboratory Assistant National Committee for Careers in Medical Technology,United States. Office of Education,1966 Guideline of a training programme for the vocational training of health service laboratory technicians in the USA includes planning administrative aspects curriculum development and equipment References

Laboratory Health and Safety Handbook R. Scott Stricoff,Douglas B. Walters,1990-09-24 Provides a comprehensive review of the components of a laboratory health and safety program applicable to commercial and academic chemical laboratories Describes how each of these components fits into a chemical hygiene plan required under OSHA s new Laboratory Use of Toxic Substances standard Features liberal use of tables and illustrations and contains a detailed bibliography

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Kerri,1997 This manual is designed to train operators in the practical hands on aspects of safely operating and maintaining small wastewater collection treatment and disposal systems It focuses on the knowledge and skills needed to operate and maintain several types of package wastewater treatment processes It also describes various effluent disposal options how to perform basic lab procedures and how to administer a small wastewater system [Answers Online](#) Barbara Newlin,1985 Understanding databases Going online the hardware and software you need Business databases Scientific databases News general information and social science databases The information utilities Electronic mail Bibliographic services After hours services database bargains Full text services and databases The numeric services Basic online searching techniques Advanced online searching techniques Document delivery Evaluating and selecting services and databases

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