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# Oedipus Rex Study Exam Questions & Answers 2025/2026

Oedipus Rex - ANSWERSKing of Thebes; murdered King Laius (his father) and married his mother (Jocasta)

Jocasta - ANSWERSKing Laius' wife; Oedipus' mother, but also wife...

Creon - ANSWERSbrother to Jocasta; acts as king after Laius is killed, and abdicates the throne when Oedipus comes.

Tiresias - ANSWERSBlind prophet Oedipus calls upon to find out who Laius' murderer is.

Antigone / Ismene - ANSWERStwo of Oedipus' children

How does Oedipus become the King of Thebes? - ANSWERSHe solves the riddle of the sphinx who has been plaguing Thebes.

Why are so many people on the steps of Oedipus' palace in the beginning of the play? - ANSWERSThere is a plague in Thebes; cattle, crops, and people are dying. (main conflict in Act 1)

Laius - ANSWERSpreviously king of Thebes, previously married to Jocasta, and murdered at a crossroads.

When Laius was murdered, why did nobody try to figure out what happened? - ANSWERSEveryone was preoccupied with the terror of the sphinx and trying to get rid of her.

# Oedipus Rex Study Questions And Answers

**Xiang Xie**



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web a write a balanced equation for the reaction Al + O<sub>2</sub> → Al<sub>2</sub>O<sub>3</sub> b determine the theoretical yield of Al<sub>2</sub>O<sub>3</sub> 160.0 g Al × 1 mol Al / 27.0 g Al × 2 mol Al<sub>2</sub>O<sub>3</sub> / 3 mol Al × 101.964 g Al<sub>2</sub>O<sub>3</sub> / 1 mol Al<sub>2</sub>O<sub>3</sub> = 302.3 g Al<sub>2</sub>O<sub>3</sub> c determine the percent yield yield actual / yield theoretical × 100 = 260.0 g / 302.3 g × 100 = 86.01%

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web 7 Na<sub>2</sub>C<sub>2</sub>O<sub>4</sub> is the limiting reactant percent yield = 86.68% only four molecules can be made 9 this amount cannot be weighted by ordinary balances and is worthless 10 nitrogen is the limiting reagent 11 yes methane is the limiting reagent 12 C is the limiting reagent 4 33 g of H<sub>2</sub> are left over 13

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