

# Determination of Available Chlorine in Bleaching Powder

## Principle

Bleaching powder is commonly used as a disinfectant. The chlorine present in the bleaching powder gets reduced with time. So, to find the exact quantity of bleaching powder required, the amount of available chlorine in the sample must be found out.

Chlorine will liberate free iodine from potassium iodide solution when its pH is 8 or less. The iodine liberated, which is equivalent to the amount of active chlorine, is titrated with standard sodium thiosulphate solution using starch as indicator.

## Bleaching powder

Bleaching powder is also referred to as T.C.L. powder (**Terephthaloyl chloride**). Bleaching powder is nothing but a compound of Lime and Chlorine.

In fresh Bleaching powder, normally chlorine content is 33 % or more. Bleaching powder is unstable in nature and as the time passes the chlorine content reduces. To avoid loss of chlorine, it is necessary to store Bleaching powder in dry condition and away from direct sun light. If the Bleaching powder is in a bag, after use, the bag should be properly closed.

- Contact period of minimum half an hour is required for the chlorine to act on the pathogens and kill them. Hence, it is necessary to do the chlorination, minimum one hour before the supply hours.
- The quantity of Bleaching powder required for disinfection depends on the quantity of water in the GSR/ESR. It can be found out with the help of a formula.

## Chlorination:

Quantity of Bleaching Powder required for disinfection as per above formula should be taken in a plastic bucket and adding small quantity of water, a thick paste should be prepared. Then the bucket should be filled up to 50% to 75% by water and the paste should be thoroughly

# Determination Of Available Chlorine In Bleaching Solution

**S Marginson**



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## **Table of Contents Determination Of Available Chlorine In Bleaching Solution**

1. Understanding the eBook Determination Of Available Chlorine In Bleaching Solution
  - The Rise of Digital Reading Determination Of Available Chlorine In Bleaching Solution
  - Advantages of eBooks Over Traditional Books
2. Identifying Determination Of Available Chlorine In Bleaching Solution
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Determination Of Available Chlorine In Bleaching Solution
  - User-Friendly Interface
4. Exploring eBook Recommendations from Determination Of Available Chlorine In Bleaching Solution
  - Personalized Recommendations
  - Determination Of Available Chlorine In Bleaching Solution User Reviews and Ratings
  - Determination Of Available Chlorine In Bleaching Solution and Bestseller Lists
5. Accessing Determination Of Available Chlorine In Bleaching Solution Free and Paid eBooks
  - Determination Of Available Chlorine In Bleaching Solution Public Domain eBooks
  - Determination Of Available Chlorine In Bleaching Solution eBook Subscription Services
  - Determination Of Available Chlorine In Bleaching Solution Budget-Friendly Options

6. Navigating Determination Of Available Chlorine In Bleaching Solution eBook Formats
  - ePub, PDF, MOBI, and More
  - Determination Of Available Chlorine In Bleaching Solution Compatibility with Devices
  - Determination Of Available Chlorine In Bleaching Solution Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Determination Of Available Chlorine In Bleaching Solution
  - Highlighting and Note-Taking Determination Of Available Chlorine In Bleaching Solution
  - Interactive Elements Determination Of Available Chlorine In Bleaching Solution
8. Staying Engaged with Determination Of Available Chlorine In Bleaching Solution
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Determination Of Available Chlorine In Bleaching Solution
9. Balancing eBooks and Physical Books Determination Of Available Chlorine In Bleaching Solution
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Determination Of Available Chlorine In Bleaching Solution
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Determination Of Available Chlorine In Bleaching Solution
  - Setting Reading Goals Determination Of Available Chlorine In Bleaching Solution
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Determination Of Available Chlorine In Bleaching Solution
  - Fact-Checking eBook Content of Determination Of Available Chlorine In Bleaching Solution
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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