

MOLAR CONVERSIONS LAB

Many everyday items are counted in groups in order to make them more manageable. Eggs are counted in dozens; sheets of paper are counted in reams. Since atoms and molecules are so minuscule, it is necessary to count them in groups much larger than 12 or 500. Instead, they are counted in groups called moles. The mole is a group of 6.02×10^{23} particles, or Avogadro's number (N_A) of particles.

Counting the individual items in a group as large as the mole is impossible. As a result, a different method is used to count a mole of particles – counting by weighing. The same process is used to count aluminum cans when you take a bag to the recycling center. A conversion factor (weight per can) is used to convert the weight of the bag into the number of cans in the bag. In the same way, the molar mass of an element or compound is used to convert the mass of a sample into the number of atoms or particles in that sample.

In this lab, you will perform molar conversions for small samples of two everyday substances, water and chalk.

PROCEDURE

1. Fill a paper cup with water and measure its mass.
2. Drink one mouthful of water and find the new mass of the paper cup.
3. Record the initial and final masses in your data table.
4. Obtain a piece of chalk (calcium carbonate). Wipe it free of dust particles and measure its mass.
5. Write your full name with the chalk and find its new mass.
6. Record the initial and final masses in your data table.

DATA – Molar Conversions Data Table

	Water	Chalk
Initial Mass (g)		
Final Mass (g)		
Mass consumed (g)		

ANALYSIS – Information for Each Substance (show work on the back of this sheet)

	Water	Chalk
Chemical Formula		
Molar Mass		
Particles Consumed in Lab		

QUESTIONS – Answer on the back of this sheet

1. Identify the two conversion factors you used and specify how you used them.
2. If you consume 5 grams each of water and chalk, will you consume more water particles or more chalk particles? Explain.
3. If you consume 5 moles each of water and chalk, will you consume more water particles or more chalk particles? Explain.

Moles Of Chalk Lab Answers

JG Myers



Moles Of Chalk Lab Answers:

E-chemistry Iii (science and Technology)' 2003 Ed. , *Laboratory Manual for Principles of General Chemistry* Jo Allan Beran,2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures **Canadian Journal of Chemistry** ,1968 *Physics Guide and Lab Activities* Ira Cleveland Davis,1957 **Prentice Hall Chemistry** ,2000 Christian Home Educators' Curriculum Manual Cathy Duffy,1997-11 *Basic Chemistry* Steven S. Zumdahl,2004 Description Not Yet Available *Nuclear Science Abstracts* ,1970 **Illinois Chemistry Teacher** ,2003 **Government Reports Announcements & Index** ,1983 **Improving Student Comprehension of Stoichiometric Concepts** Connie Lynn Bannick Kemner,2007 *Biology* Eric Strauss,Marylin Lisowski,2000 **Canadian Chemical Processing** ,1969-07

Moles Of Chalk Lab Answers: Bestsellers in 2023 The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous engrossing novels captivating the hearts of readers worldwide. Lets delve into the realm of popular books, exploring the fascinating narratives that have enthralled audiences this year. Moles Of Chalk Lab Answers : Colleen Hoover "It Ends with Us" This poignant tale of love, loss, and resilience has gripped readers with its raw and emotional exploration of domestic abuse. Hoover skillfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can prevail. Moles Of Chalk Lab Answers : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This captivating historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids compelling storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Moles Of Chalk Lab Answers : Delia Owens "Where the Crawdads Sing" This evocative coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens weaves a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These popular novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of compelling stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a masterful and thrilling novel that will keep you guessing until the very end. The novel is a warning tale about the dangers of obsession and the power of evil.

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