

Science

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(Chapter - 1) (Chemical Reactions and Equations) (Exercises) (Class 10)

Question 14:

In the refining of silver, the recovery of silver from silver nitrate solution involved displacement by copper metal. Write down the reaction involved.

Answer 14:



Question 15:

What do you mean by a precipitation reaction? Explain by giving examples.

Answer 15:

A reaction in which an insoluble solid (called precipitate) is formed is called a precipitation reaction.



In this reaction, barium sulphate is obtained as a precipitate. Hence, it is a precipitation reaction.

Question 16:

Explain the following in terms of gain or loss of oxygen with two examples each.

(a) Oxidation

(b) Reduction

Answer 16:

(a) Oxidation: Oxidation is the gain of oxygen.



Here, carbon (C) is oxidised to carbon dioxide (CO₂) and copper (Cu) is oxidised to copper oxide (CuO).

(b) Reduction: Reduction is the loss of oxygen.



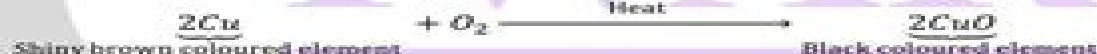
Here, carbon dioxide (CO₂) is reduced to carbon monoxide (CO) and copper oxide (CuO) is reduced to copper (Cu).

Question 17:

A shiny brown coloured element 'X' on heating in air becomes black in colour. Name the element 'X' and the black coloured compound formed.

Answer 17:

The element 'X' is copper (Cu) which is converted into black coloured copper oxide (CuO).



Question 18:

Why do we apply paint on iron articles?

Answer 18:

Iron articles are painted because applying paints prevents them from rusting. When painted, the contact of iron articles from moisture and air is cut off. Hence, rusting is prevented their presence is essential for rusting to take place.

Chapter 9 Chemical Reactions Answers

Patrick Vollmar



Chapter 9 Chemical Reactions Answers:

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Table of Contents Chapter 9 Chemical Reactions Answers

1. Understanding the eBook Chapter 9 Chemical Reactions Answers
 - The Rise of Digital Reading Chapter 9 Chemical Reactions Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 9 Chemical Reactions Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter 9 Chemical Reactions Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 9 Chemical Reactions Answers
 - Personalized Recommendations
 - Chapter 9 Chemical Reactions Answers User Reviews and Ratings

- Chapter 9 Chemical Reactions Answers and Bestseller Lists
- 5. Accessing Chapter 9 Chemical Reactions Answers Free and Paid eBooks
 - Chapter 9 Chemical Reactions Answers Public Domain eBooks
 - Chapter 9 Chemical Reactions Answers eBook Subscription Services
 - Chapter 9 Chemical Reactions Answers Budget-Friendly Options
- 6. Navigating Chapter 9 Chemical Reactions Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chapter 9 Chemical Reactions Answers Compatibility with Devices
 - Chapter 9 Chemical Reactions Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chapter 9 Chemical Reactions Answers
 - Highlighting and Note-Taking Chapter 9 Chemical Reactions Answers
 - Interactive Elements Chapter 9 Chemical Reactions Answers
- 8. Staying Engaged with Chapter 9 Chemical Reactions Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter 9 Chemical Reactions Answers
- 9. Balancing eBooks and Physical Books Chapter 9 Chemical Reactions Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter 9 Chemical Reactions Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chapter 9 Chemical Reactions Answers
 - Setting Reading Goals Chapter 9 Chemical Reactions Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chapter 9 Chemical Reactions Answers
 - Fact-Checking eBook Content of Chapter 9 Chemical Reactions Answers
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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