



## The Chemistry of Life

## Chapter Test A

### Multiple Choice

Directions: Select the best answer for each of the questions by circling the number of the best possible answer.

1. If an atom has a mass number of 23 and 11 electrons, the atom is most likely to
  - a. be neutral
  - b. be positive
  - c. be negative
  - d. be an ion
2. If an atom has a mass number of 23 and 11 protons, the atom is most likely to
  - a. be neutral
  - b. be positive
  - c. be negative
  - d. be an ion
3. Which of the following is not a constituent of water?
  - a. two atoms of hydrogen and one atom of oxygen
  - b. one atom of hydrogen and one atom of oxygen
  - c. one atom of hydrogen and two atoms of oxygen
  - d. two atoms of hydrogen and two atoms of oxygen
4. A molecule that is neutral is the molecule
  - a. containing electrons
  - b. containing protons
  - c. having electrons
  - d. having protons
5. Which molecule is polar, with the
  - a. oxygen with being slightly positive and the hydrogen with being slightly negative
  - b. oxygen with hydrogen with being slightly positive
  - c. oxygen with hydrogen with being slightly negative
  - d. oxygen with being slightly negative and the hydrogen with being slightly positive
6. If you are with water, which water are you in?
  - a. positive relative humidity
  - b. relative wetness
  - c. relative humidity
  - d. relative wet
7. A substance with a pH of 7 is called
  - a. an acid
  - b. basic
  - c. neutral
  - d. a compound
8. Relative humidity is 100% if
  - a. it rains
  - b. it snows
  - c. it rains
  - d. it rains
9. Why is coffee considered compared to other drinks?
  - a. Coffee does not contain any water and thus is not a different substance.
  - b. Coffee does not have quality taste.
  - c. With coffee there are three molecules made with oxygen and hydrogen.
  - d. With coffee there are the molecules of water, caffeine, and compounds.

# Chapter Test Chemistry Of Life Answer Key

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