

# The Chemistry of Cookies

Get this "bake" gone with the chemistry lab. Students measure the heat of the oven and the amount of time needed to bake cookies and then compare the results to the amount of time needed to bake cookies. This will be a fun and educational activity for students. (1001-1001)

**Table 1: Empirical Molecular Formula** (from the following procedure for the calculation of the empirical molecular formula) (1001-1001)

Amount	Chemical Formula	Amount of Carbon	Amount of Hydrogen
Baking Soda	$\text{NaHCO}_3$	12.01 g	1.01 g
Baking Powder	$\text{NaHCO}_3$	12.01 g	1.01 g
Baking Sugar	$\text{C}_{12}\text{H}_{22}\text{O}_{11}$	36.03 g	22.02 g
Butter	$\text{C}_{55}\text{H}_{98}\text{O}_6$	60.09 g	98.09 g
Vanilla Sugar	$\text{C}_{10}\text{H}_{12}\text{O}_4$	12.01 g	12.01 g
Chocolate Chips	$\text{C}_{10}\text{H}_{12}\text{O}_4$	12.01 g	12.01 g
Confectioner	$\text{C}_{12}\text{H}_{22}\text{O}_{11}$	36.03 g	22.02 g
Sugar	$\text{C}_{12}\text{H}_{22}\text{O}_{11}$	36.03 g	22.02 g

**Table 2: Cookie Ingredients** (from the following procedure for the calculation of the empirical molecular formula) (1001-1001)

- 1 cup of baking soda = 250g
- 1 cup of vanilla sugar = 125g
- 1 cup of sugar = 200g
- 1 cup of butter = 225g
- 1 cup of confectioner = 125g
- 1 cup of sugar = 200g
- 1 cup of sugar = 200g

## Ingredients

1 cup of vanilla sugar (125g) and 1 cup of confectioner (125g) (1001-1001)

1 cup of sugar (200g) and 1 cup of butter (225g) (1001-1001)

1 cup of baking soda (250g) and 1 cup of vanilla sugar (125g) (1001-1001)

1 cup of sugar (200g) and 1 cup of confectioner (125g) (1001-1001)

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# Chemistry Chocolate Chip Cookies Answers

**Robert R. Redfield**



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