

3 • Reactions in Aqueous Solution**STUDY QUESTIONS AND PROBLEMS**

1. Classify each of the following solutes as a strong electrolyte, weak electrolyte, or nonelectrolyte:

sugar	sodium hydroxide
common salt (NaCl)	hydrochloric acid
alcohol	copper sulfate
acetic acid	carbonic acid

2. Predict the solubility of the following salts:

- sodium sulfate
- potassium chromate
- silver bromide
- nickel (II) hydroxide
- aluminum nitrate
- barium sulfide
- ammonium acetate
- strontium iodide

3. Write the ions that are produced when the following substances dissolve in water:

- $Mg(OH)_2$
- K_2SO_4
- $NaHCO_3$
- $(NH_4)_3PO_4$
- $NaClO$

4. Predict whether or not the following reactions will lead to a precipitate. Write complete and net ionic equations for all the reactions.

- a. potassium chromate and lead acetate
- b. silver perchlorate and ammonium chloride
- c. potassium carbonate and copper acetate
- d. sodium fluoride and magnesium iodide
- e. barium nitrate and potassium sulfate

Reactions In Aqueous Solution Worksheet Answers

Michael Brown



Reactions In Aqueous Solution Worksheet Answers:

Reviewing **Reactions In Aqueous Solution Worksheet Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Reactions In Aqueous Solution Worksheet Answers**," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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