

Chapter 4 – Atomic Structure

Short Answer: Answer the following questions in complete sentences using science vocabulary. Make sure that your answers are complete and detailed.

1. A sample of calcium contains calcium-40, calcium-44, calcium-42, calcium-48, calcium-43, and calcium-46 atoms. Explain why these atoms have different mass numbers but must have the same atomic number.

The atoms have different mass numbers because they have a different number of neutrons. The mass number is equaled to the number of protons plus neutrons.

The atoms must have the same atomic number because the atomic number is equaled to the number of protons. The number of protons is unique for each element.

2. Complete the following table. Circle the name of the element(s) that are isotopes. Complete Bohr's Model diagrams for the element with eleven electrons and the element with an atomic number of 9.

Element Name	Element Symbol	Atomic Number	Mass Number	Protons	Neutrons	Electrons
Scandium	Sc	21	45	21	24	21
Fluorine	F	9	19	9	10	9
Sodium	Na	11	24	11	13	11
Answers Vary					82	

Chapter 4 Atomic Structure Practice Problems Answers

JL Elias



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Enjoying the Beat of Term: An Emotional Symphony within **Chapter 4 Atomic Structure Practice Problems Answers**

In a world used by screens and the ceaseless chatter of instantaneous communication, the melodic beauty and psychological symphony created by the written term frequently diminish in to the backdrop, eclipsed by the persistent noise and interruptions that permeate our lives. Nevertheless, nestled within the pages of **Chapter 4 Atomic Structure Practice Problems Answers** a wonderful fictional prize overflowing with raw feelings, lies an immersive symphony waiting to be embraced. Constructed by an outstanding musician of language, that captivating masterpiece conducts visitors on a mental journey, skillfully unraveling the hidden tunes and profound impact resonating within each cautiously crafted phrase. Within the depths of this touching evaluation, we shall examine the book is main harmonies, analyze their enthralling publishing fashion, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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