

**CHAPTER 7 REVIEW***Chemical Formulas and Chemical Compounds***SECTION 7-1****SHORT ANSWER** Answer the following questions in the space provided.

1. C In a Stock name such as iron(III) sulfate, the roman numeral tells us \_\_\_\_\_
- how many atoms of Fe are in one formula unit
  - how many sulfate ions can be attached to the iron atom
  - the charge on each Fe ion
  - the total positive charge of the formula unit
2. C The result of changing a subscript in a correctly written chemical formula is to \_\_\_\_\_
- change the number of moles represented by the formula
  - change the charges on the other ions in the compound
  - change the formula so that it no longer represents the compound it previously represented
  - have no effect on the formula
3. The explosive TNT has the molecular formula  $C_7H_5(NO_2)_3$ .
- |  |   |
|--|---|
| <u>4 elements</u>                            | a. How many elements make up this compound?   |
| <u>6 oxygen atoms</u>                        | b. How many oxygen atoms are present in one molecule of $C_7H_5(NO_2)_3$ ?                      |
| <u>21 atoms</u>                              | c. How many atoms in total are present in one molecule of $C_7H_5(NO_2)_3$ ?                    |
| <u><math>4.2 \times 10^{24}</math> atoms</u> | d. How many atoms are present in a sample of $2 \times 10^{23}$ molecules of $C_7H_5(NO_2)_3$ ? |
4. How many atoms are present in each of these formula units?
- |                 |                         |
|-----------------|-------------------------|
| <u>11 atoms</u> | a. $Ca(HCO_3)_2$        |
| <u>45 atoms</u> | b. $C_{12}H_{22}O_{11}$ |
| <u>10 atoms</u> | c. $Fe(ClO_2)_3$        |
| <u>9 atoms</u>  | d. $Fe(ClO_3)_3$        |
5.  $N_2O_5$  a. What is the formula for the compound dinitrogen pentoxide?
- carbon(IV) sulfide b. What is the Stock name for the covalent compound  $CS_2$ ?
- $H_2SO_3$  c. What is the formula for sulfurous acid?
- phosphoric acid d. What is the name for the acid  $H_3PO_4$ ?

# Chapter 7 Review Chemical Formulas And Chemical Compounds

**J Elliott**



## **Chapter 7 Review Chemical Formulas And Chemical Compounds:**

## Whispering the Techniques of Language: An Psychological Journey through **Chapter 7 Review Chemical Formulas And Chemical Compounds**

In a digitally-driven earth where screens reign great and instant transmission drowns out the subtleties of language, the profound strategies and mental subtleties hidden within words frequently go unheard. Yet, set within the pages of **Chapter 7 Review Chemical Formulas And Chemical Compounds** a charming literary treasure pulsing with fresh feelings, lies an extraordinary journey waiting to be undertaken. Written by an experienced wordsmith, this marvelous opus invites visitors on an introspective trip, softly unraveling the veiled truths and profound influence resonating within ab muscles material of each word. Within the emotional depths of the poignant review, we can embark upon a heartfelt exploration of the book is core styles, dissect its captivating writing model, and fail to the effective resonance it evokes serious within the recesses of readers hearts.

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