***Stoichiometry Worksheet 2***
***Chemistry***

Name_____

Date_____ Hour_____

1. In the reaction : $2Al_2O_{3(l)} \rightarrow 4Al_{(s)} + 3O_{2(g)}$, the mole ratios are:

$Al_2O_3$ to Al _____ $Al_2O_3$ to $O_2$ _____ Al to $O_2$ _____ $O_2$ to Al _____

2. In the reaction: $2H_2 + O_2 \rightarrow 2H_2O$, the mole ratio of oxygen to water is: _____

3. In the reaction: $Ca + Cl_2 \rightarrow CaCl_2$, the mole ratio of chlorine to calcium is: _____

4. In the reaction: $Zn + H_2SO_4 \rightarrow ZnSO_4 + H_2$, the mole ratio of zinc to sulfuric acid is:

_____

5. The equation for the Haber process is represented by:

$$N_{2(g)} + 3H_{2(g)} \rightarrow 2NH_{3(g)}$$

To convert 9.0 moles of hydrogen to ammonia would require how many moles of nitrogen?

6. In the equation: $2KClO_{3(g)} \rightarrow 2KCl_{(s)} + 3O_{2(g)}$, how many moles of oxygen are produced when 3.00 moles of $KClO_3$ decompose completely?

7. For the reaction: $C_{(s)} + 2H_{2(g)} \rightarrow CH_{4(g)}$, how many moles of $H_2$ are required to produce 10.0 moles of $CH_4$?

# Chapter 9 Stoichiometry Section 2 Worksheet

Leo J. Malone

**Chapter 9 Stoichiometry Section 2 Worksheet:**

**Stoichiometry and Thermodynamics of Metallurgical Processes** Y. K. Rao,1985-10-31 Originally published in 1985 this textbook provides a thorough and comprehensive coverage of a wide range of topics in stoichiometry and thermodynamics with special emphasis on applications to metallurgical processes This book will be welcomed as a text for courses in elementary and advanced thermodynamics and stoichiometry **MCAT Comprehensive Review 1999** ,1998-11 Complete Review of ALL the science you need to know Full Length Practice Test with complete explanations for each question 2 Full Length Verbal Reasoning Tests with detailed explanations and strategies **Handbook on Material and Energy Balance Calculations in Material Processing, Includes CD-ROM** Arthur E. Morris,H. Alan Fine,Gordon Harold Geiger,2011-09-06 This book approaches the subject of material and energy balances from two directions First it emphasizes the fundamental principles of the conservation of mass and energy and the consequences of these two principles Second it applies the techniques of computational chemistry to materials processing and introduces new software developed by the author especially for material and heat balances The third edition reflects the changes in the professional engineer s practice in the last 30 years reflecting the dramatic shift away from metallurgical engineering and the extractive industry towards materials engineering A large and growing number of recent graduates are employed in such fields as semiconductor processing environmental engineering and the production and processing of advanced and exotic materials for aerospace electronic and structural applications The advance in computing power and software for the desktop computer has significantly changed the way engineers make computations and the biggest change comes from the computational approach used to solve problems The spreadsheet program Excel is used extensively throughout the text as the main computational engine for solving material and energy balance equations and for statistical analysis of data The use of Excel and the introduction of the add in programs enables the study of a range of variables on critical process parameters and emphasis is placed on multi device flowsheets with recycle bypass and purge streams whose material and heat balance equations were previously too complicated to solve by the normally used hand calculator The Excel based program FlowBal helps the user set up material and heat balance equations for processes with multiple streams and units **Holt Chemistry** Ralph Thomas Myers,2004 **MCAT Comprehensive Review, 2004** Kaplan, Inc,2003 The bestselling MCAT prep guide is a must for pre med students preparing for the demanding Medical College Admissions Test MCAT The CD features one full length practice test with detailed explanations plus effective tips and strategies **MCAT Comprehensive Review** ,2004 *Basic Concepts of Chemistry* Leo J. Malone,Theodore O. Dolter,2011-12-27 The 9th edition of Malone s Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the

end of each chapter Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter WileyPLUS sold separately from text Chemistry Workbook For Dummies Peter J. Mikulecky,Katherine Brutlag,Michelle Rose Gilman,Brian Peterson,2008-08-06 From liquids and solids to acids and bases work chemistry equations and use formulas with ease Got a grasp on the chemistry terms and concepts you need to know but get lost halfway through a problem or worse yet not know where to begin Have no fear this hands on guide helps you solve many types of chemistry problems in a focused step by step manner With problem solving shortcuts and lots of practice exercises you ll build your chemistry skills and improve your performance both in and out of the science lab You ll see how to work with numbers atoms and elements make and remake compounds understand changes in terms of energy make sense of organic chemistry and more 100s of Problems Know where to begin and how to solve the most common chemistry problems Step by step answer sets clearly identify where you went wrong or right with a problem Understand the key exceptions to chemistry rules Use chemistry in practical applications with confidence *CliffsNotes AP Chemistry 2021 Exam* Angela Woodward Spangenberg,2020-09-29 CliffsNotes AP Chemistry 2021 Exam gives you exactly what you need to score a 5 on the exam concise chapter reviews on every AP Chemistry subject in depth laboratory investigations and full length model practice exams to prepare you for the May 2021 exam Revised to even better reflect the new AP Chemistry exam this test prep guide includes updated content tailored to the May 2021 exam Features of the guide focus on what AP Chemistry test takers need to score high on the exam Reviews of all subject areas In depth coverage of the all important laboratory investigations Two full length model practice AP Chemistry exams Every review chapter includes review questions and answers to pinpoint problem areas **Student Study Guide Fundamentals of Chemistry by David E. Goldberg** Margaret G. Kimble,1994 *The Practice of Chemistry* Donald J. Wink,Sharon Fetzer-Gislason,Sheila McNicholas,2003-03 Students can t do chemistry if they can t do the math The Practice of Chemistry First Edition is the only preparatory chemistry text to offer students targeted consistent mathematical support to make sure they understand how to use math especially algebra in chemical problem solving The book s unique focus on actual chemical practice extensive study tools and integrated media makes The Practice of Chemistry the most effective way to prepare students for the standard general chemistry course and bright futures as science majors This special PowerPoint tour of the text was created by Don Wink http www bfwpub com pdfs wink POCPowerPoint_Final ppt 832KB **Chemical Problem-solving by Dimensional Analysis** Arnold B. Loebel,1978 **Instructor's Manual and Test Bank to Accompany Basic Concepts of Chemistry** Malone,2000-04-18 *Basic Concepts of Chemistry, Study Guide* Leo J. Malone,2000 The text s three main goals are to introduce chemistry as a living relevant science to encourage learning and

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