

20 Points

Name: Phyllis A. Smith Lab Section: 200

Suppose that you had the correct equation for the following combustion reaction. What highest boiling point compound?

$\text{C}_2\text{H}_6\text{O}$

C_2H_6

C_2H_4

C_2H_2

$\text{C}_2\text{H}_6\text{O}$
Ethanol

Diethyl ether

C_2H_4

C_2H_2
Acetylene

1. If you have the same conditions of a 1.00 g sample of C_2H_6 or $\text{C}_2\text{H}_6\text{O}$, which would be completely converted to water to yield a 1.00 g sample of H_2O ?



2.34 mol

2. If you were in the market and wanted the same amount of C_2H_6 or $\text{C}_2\text{H}_6\text{O}$ in the amount of the water in 1.00 g of H_2O ?



0.567 mol

0.567 mol

0.567 mol

0.567 mol

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Richard Bailey



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