

Lab 22A

Investigating Chemical Equilibrium

Purpose: To investigate the macroscopic properties of two chemical systems at equilibrium, observe shifts in equilibrium concentrations as a result of concentration and temperature and relate the to systems, LeChatelier's principle, and equilibrium through applying Le Chatelier's principle.

Proble:

1. Write a balanced equation for the reaction in part 1.
$$\text{Fe}^{3+}(\text{aq}) + \text{SCN}^{-}(\text{aq}) \rightleftharpoons \text{FeSCN}^{2+}(\text{aq})$$
2. Write a balanced equation for the reaction in part 2.
$$\text{Fe}^{3+}(\text{aq}) + \text{SCN}^{-}(\text{aq}) \rightleftharpoons \text{FeSCN}^{2+}(\text{aq})$$
3. Write a balanced equation for the reaction in part 3.
$$\text{Fe}^{3+}(\text{aq}) + \text{SCN}^{-}(\text{aq}) \rightleftharpoons \text{FeSCN}^{2+}(\text{aq})$$
4. In part 1, what is the conjugate acid and base? In part 2, what is the conjugate acid and base?
In part 1, the conjugate acid is FeSCN^{2+} and the conjugate base is Fe^{3+} . In part 2, the conjugate acid is FeSCN^{2+} and the conjugate base is Fe^{3+} .
5. What is the conjugate acid and base in part 3?
In part 3, the conjugate acid is FeSCN^{2+} and the conjugate base is Fe^{3+} .
6. Write the equation for the reaction between Fe^{3+} and OH^{-} ions.
$$\text{Fe}^{3+}(\text{aq}) + \text{OH}^{-}(\text{aq}) \rightleftharpoons \text{Fe}(\text{OH})^{2+}(\text{aq})$$
7. Write the equation for the reaction between Fe^{3+} and OH^{-} ions.
$$\text{Fe}^{3+}(\text{aq}) + \text{OH}^{-}(\text{aq}) \rightleftharpoons \text{Fe}(\text{OH})^{2+}(\text{aq})$$
8. What is the conjugate acid and base in part 6?
In part 6, the conjugate acid is $\text{Fe}(\text{OH})^{2+}$ and the conjugate base is Fe^{3+} .
9. Write the equation for the reaction between Fe^{3+} and OH^{-} ions.
$$\text{Fe}^{3+}(\text{aq}) + \text{OH}^{-}(\text{aq}) \rightleftharpoons \text{Fe}(\text{OH})^{2+}(\text{aq})$$
10. Write the equation for the reaction between Fe^{3+} and OH^{-} ions.
$$\text{Fe}^{3+}(\text{aq}) + \text{OH}^{-}(\text{aq}) \rightleftharpoons \text{Fe}(\text{OH})^{2+}(\text{aq})$$

Investigating Chemical Equilibrium Lab 12a Answer

Jiyuan Zhang



Investigating Chemical Equilibrium Lab 12a Answer:

Investigating Chemical Equilibrium Lab 12a Answer Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the energy of words has been evident than ever. They have the ability to inspire, provoke, and ignite change. Such is the essence of the book **Investigating Chemical Equilibrium Lab 12a Answer**, a literary masterpiece that delves deep into the significance of words and their effect on our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book's key themes, examine its writing style, and analyze its overall impact on readers.

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