

8E Combustion- Revision Worksheet

Define the following keywords:

Combustion	
Fuel	
Fossil Fuel	
Hydrocarbon	
Reactant	
Product	

Describe how hydrogen can be used to power vehicles:

.....

.....

Write out a word equation for the reaction in a fuel cell:

Describe the energy transfers that occur during combustion.

.....

.....

State the products of hydrocarbon combustion and explain how they are formed.

.....

.....

.....

.....

Explain how the equipment here allows you to test for the products of combustion.

.....

.....

Complete the following word equations:

Magnesium + Oxygen →

Iron + Oxygen →

Copper + Oxygen →

Zinc + Oxygen →

Explain why the reactions above are examples of oxidation.

.....

.....

Describe what is meant by the 'law of conservation of mass'

.....

.....

5g of magnesium is burnt in air, producing 10g of magnesium oxide. Calculate how much oxygen was reacted with.

Describe the reaction of magnesium with oxygen.

Explain why the mass of zinc oxide is greater than the mass of the zinc reacted.

.....

Describe why combustion is an exothermic reaction.

.....

.....

Label the fire triangle:

State what each of the hazard symbols means:

Complete the following table about the different fire extinguishers:

Type	Colour	Explanation of what it puts out
Water		
CO2		
Foam		
Powder		

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KJ Lindholm-Leary



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