

CONFORMATIONAL ANALYSIS: PRACTICE EXERCISES

10. Draw a Newman projection of the most stable conformation of 2-methylpropane.

11. The structures below are



- A) ethane.
- B) conformational isomers.
- C) structural isomers.
- D) both B and C.

12. Define the term conformation.

13. Draw a Newman molecule along the C_2-C_3 bond and provide a Newman projection of the lowest energy conformer.

14. Provide a representation of the gauche conformer of butane.

15. Among the butane conformers, which occur at energy minima on a graph of potential energy versus dihedral angle?

- A) gauche only
- B) eclipsed and totally eclipsed
- C) anti and anti
- D) eclipsed only
- E) anti only

16. Which of the following substituents exhibits the greatest negative heat of combustion per CH_2 group?

- A) cyclohexane B) cyclopentane C) cyclohexene D) cyclohexane E) cyclopentane

17. Which of the following correctly ranks the substituents in order of increasing the heat per methylene (CH_2) group?

- A) cyclopentane < cyclohexane < cyclohexene < cyclohexane
- B) cyclohexane < cyclopentane < cyclohexene < cyclopentane
- C) cyclohexane < cyclohexene < cyclopentane < cyclopentane
- D) cyclopentane < cyclopentane < cyclohexane < cyclohexane
- E) cyclopentane < cyclohexane < cyclohexene < cyclohexane

18. Describe the number of angles which will be formed when placed in cyclopentane.

Conformational Analysis Practice Exercises

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