

## CHAPTER 7 REVIEW

### *Chemical Formulas and Chemical Compounds*

#### MIXED REVIEW

**SHORT ANSWER** Answer the following questions in the space provided.

1. Write formulas for the following compounds:

- \_\_\_\_\_ a. copper(II) carbonate  
 \_\_\_\_\_ b. sodium sulfite  
 \_\_\_\_\_ c. ammonium phosphate  
 \_\_\_\_\_ d. tin(IV) sulfide  
 \_\_\_\_\_ e. nitrous acid

2. Write the Stock names for the following compounds:

- \_\_\_\_\_ a.  $\text{Mg}(\text{ClO}_4)_2$   
 \_\_\_\_\_ b.  $\text{Fe}(\text{NO}_3)_2$   
 \_\_\_\_\_ c.  $\text{Fe}(\text{NO}_2)_3$   
 \_\_\_\_\_ d.  $\text{CoO}$   
 \_\_\_\_\_ e. dinitrogen pentoxide

3. \_\_\_\_\_ a. How many atoms are represented by the formula  $\text{Ca}(\text{HSO}_4)_2$ ?  
 \_\_\_\_\_ b. How many moles of oxygen atoms are in a 0.50 mol sample of this compound?  
 \_\_\_\_\_ c. Assign the oxidation number to sulfur in the  $\text{HSO}_4^-$  anion.

4. Assign the oxidation number to the element specified in each of the following:

- \_\_\_\_\_ a. hydrogen in  $\text{H}_2\text{O}_2$   
 \_\_\_\_\_ b. hydrogen in  $\text{MgH}_2$   
 \_\_\_\_\_ c. sulfur in  $\text{S}_8$   
 \_\_\_\_\_ d. carbon in  $(\text{CO}_3)^{2-}$   
 \_\_\_\_\_ e. chromium in  $\text{Na}_2\text{Cr}_2\text{O}_7$   
 \_\_\_\_\_ f. nitrogen in  $\text{NO}_2$

# Chapter 7 3 Answers Chemical Formulas And Chemical Compounds

**CO Houle**



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web what does the crossed arrow represent in the comic strip 11 what are two of the definitions of dipole given in the comic strip 12 making sense 13 if you finish early what does electronegativity have to do with polarity using polar bears and penguins create an illustration showing a hydrogen sulfide molecule h<sub>2</sub>s

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web each dipole represents one polar bond the molecule is nonpolar 10 describe what you think is happening to the penguins in the illustration of h<sub>2</sub>o on page 4 they are being pulled towards the stronger polar bear he s more electronegative 11 explain what you think the crossed arrow represents in the comic book polar bonds perhaps 12

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web answer because the two polar bears or penguins have equal electronegativity they will pull electrons equally hard and in most cases neither will end up with more electrons each polar bear and penguin has an electronegativity of the same value a non polar bond is formed when two polar bears or two penguins are bonded together

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web to answer this question you will examine 1 electronegativity 2 nonpolar vs polar molecules 3 electronegativity and bonding electronegativity the h atom and the cl atom in hydrogen chloride

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web so the final representation of the nh<sub>3</sub> molecule using polar bears and penguins is as follows begin array c text polar bear nitrogen text with two lone pairs of electrons text and three penguins hydrogen end array this is a fun and creative way to represent the nh<sub>3</sub> molecule

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web polar bears and penguins purpose in this lesson you will be exploring polarity and bonding between atoms in greater detail a comic book will provide new information about these topics and will introduce you to the concept of electronegativity which helps us understand partial charges

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