

**CHAPTER 7 REVIEW***Chemical Formulas and Chemical Compounds***MIXED REVIEW**

**SHORT ANSWER** Answer the following questions in the space provided.

1. Write formulas for the following compounds:

- \_\_\_\_\_ a. copper(II) carbonate  
\_\_\_\_\_ b. sodium sulfite  
\_\_\_\_\_ c. ammonium phosphate  
\_\_\_\_\_ d. tin(IV) sulfide  
\_\_\_\_\_ e. nitrous acid

2. Write the Stock names for the following compounds:

- \_\_\_\_\_ a.  $\text{Mg}(\text{ClO}_4)_2$   
\_\_\_\_\_ b.  $\text{Fe}(\text{NO}_3)_2$   
\_\_\_\_\_ c.  $\text{Fe}(\text{NO}_2)_3$   
\_\_\_\_\_ d.  $\text{CoO}$   
\_\_\_\_\_ e. dinitrogen pentoxide

3. \_\_\_\_\_ a. How many atoms are represented by the formula  $\text{Ca}(\text{HSO}_4)_2$ ?  
\_\_\_\_\_ b. How many moles of oxygen atoms are in a 0.50 mol sample of this compound?  
\_\_\_\_\_ c. Assign the oxidation number to sulfur in the  $\text{HSO}_4^-$  anion.

4. Assign the oxidation number to the element specified in each of the following:

- \_\_\_\_\_ a. hydrogen in  $\text{H}_2\text{O}_2$   
\_\_\_\_\_ b. hydrogen in  $\text{MgH}_2$   
\_\_\_\_\_ c. sulfur in  $\text{S}_8$   
\_\_\_\_\_ d. carbon in  $(\text{CO}_3)^{2-}$   
\_\_\_\_\_ e. chromium in  $\text{Na}_2\text{Cr}_2\text{O}_7$   
\_\_\_\_\_ f. nitrogen in  $\text{NO}_2$

# Chapter 7 Review Chemical Formulas And Compounds Answer Key

**Charles H. Corwin**



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