

$$\text{a) Rate} = \frac{-1}{2} \frac{d[\text{NO}]}{dt} = \frac{-d[\text{O}_2]}{dt} = \frac{1}{2} \frac{d[\text{NO}_2]}{dt}$$

$$\text{b) } \frac{-d[\text{O}_2]}{dt} = \frac{1}{2} \frac{d[\text{NO}_2]}{dt}$$

$$\frac{d[\text{NO}_2]}{dt} = 2 \times \left(\frac{-d[\text{O}_2]}{dt} \right) = 2 \times 0.2 \text{ mol L}^{-1} \text{ s}^{-1}$$

$$= 0.4 \text{ mol L}^{-1} \text{ s}^{-1}$$

Chemistry Chemical Kinetics Problems Answers

Lincoln University (Jefferson City, Mo.)

Chemistry Chemical Kinetics Problems Answers:

Unveiling the Power of Verbal Artistry: An Psychological Sojourn through **Chemistry Chemical Kinetics Problems Answers**

In a global inundated with screens and the cacophony of immediate connection, the profound power and mental resonance of verbal artistry frequently disappear in to obscurity, eclipsed by the continuous onslaught of sound and distractions. However, situated within the lyrical pages of **Chemistry Chemical Kinetics Problems Answers**, a interesting work of fictional elegance that impulses with natural thoughts, lies an memorable journey waiting to be embarked upon. Composed by a virtuoso wordsmith, this interesting opus courses viewers on an emotional odyssey, gently revealing the latent possible and profound influence embedded within the elaborate web of language. Within the heart-wrenching expanse with this evocative evaluation, we can embark upon an introspective exploration of the book is central themes, dissect its captivating publishing design, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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