



Equilibrium

- **Equilibrium** is a state where the net force on an object is zero. It can be static or dynamic.
- **Static Equilibrium**: The object is at rest.
 - $\sum F_x = 0$ and $\sum F_y = 0$
 - $\sum \tau = 0$
- **Dynamic Equilibrium**: The object is moving with constant velocity.
 - $\sum F_x = 0$ and $\sum F_y = 0$
 - $\sum \tau = 0$
- **Center of Mass**: The point where the mass of the object is concentrated.
 - $\sum \tau = 0$ if the center of mass is directly above the pivot.
- **Stable Equilibrium**: The object returns to its original position after a small displacement.
- **Unstable Equilibrium**: The object moves away from its original position after a small displacement.
- **Neutral Equilibrium**: The object remains in its original position after a small displacement.



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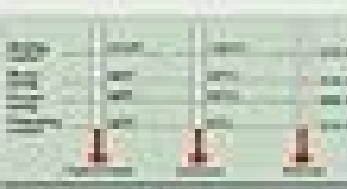
Nuclear Chemistry

- **Nuclear Chemistry** is the study of the changes in the nucleus of an atom. It involves the emission of particles and energy.
- **Radioactivity**: The process by which an unstable nucleus loses energy by emitting ionizing particles.
- **Types of Radiation**:
 - **Alpha**: α particles (Helium nuclei)
 - **Beta**: β particles (Electrons or Positrons)
 - **Gamma**: γ rays (High energy photons)
- **Half-life**: The time taken for the activity of a substance to decrease to half its initial value.
- **Radioactive Decay**: The process by which a radioactive substance loses its radioactivity over time.
- **Applications**:
 - **Medicine**: Radioisotopes are used for diagnosis and treatment.
 - **Industry**: Radioisotopes are used for tracing and dating.
 - **Energy**: Nuclear fission and fusion are used for power generation.



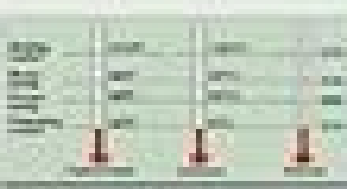
Measurement & Units

- **Measurement** is the process of determining the magnitude of a physical quantity.
- **Units**: The standard units used for measurement.
 - **SI Units**: International System of Units.
 - **Base Units**: Length (m), Mass (kg), Time (s), Temperature (K), Amount of substance (mol), Luminous intensity (cd).
 - **Derived Units**: Force (N), Energy (J), Power (W), etc.
- **Significant Figures**: The digits in a number that carry meaning contributing to its precision.
- **Accuracy**: The closeness of a measurement to the true value.
- **Precision**: The reproducibility of a measurement.



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H Kauffman



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