

Formula	Lewis Structure	Shape	Bond Angle	Dipole (yes/no)
7 CH <sub>3</sub> OH	 O: 24 V: 14	tetrahedron	109.5°	Yes
8 SO <sub>2</sub> Cl <sub>2</sub>	 O: 40 V: 32	Tetrahedron	109.5°	Yes
9 OCl <sub>2</sub>	 O: 24 V: 20	Bent	109.5°	Yes
10 SO <sub>2</sub>	 O: 24 V: 18	Bent	120°	Yes
11 SO <sub>3</sub>	 O: 32 V: 24	Trigonal Planar	120°	No
12 ClO <sub>3</sub> <sup>-</sup>	 O: 32 V: 26	Trigonal Pyramid	109.5°	Yes
13 N <sub>2</sub> O	 O: 24 V: 16	Linear	180°	Yes
O <sub>3</sub>		Bent	120°	Yes

# Molecular Geometry Lab With Answer Key

**Camilla Rothe**



**Molecular Geometry Lab With Answer Key:**

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