

NCERT Solution for Class 12  
Chemistry  
Chapter 3 - Chemical Kinetics

Intext Exercise

1. For the reaction  $R \rightarrow P$ , the concentration of a reactant changes from 0.03 M to 0.02 M in 25 minutes. Calculate the average rate of reaction using units of time both in minutes and seconds.

Ans: The average rate of the reaction can be calculated by dividing the change in the rate of decreasing the rate of reactant by the time taken. This is given below:

$$\text{Average rate} = - \frac{d[R]}{dt}$$

This can be written as:

$$\text{Average rate} = - \frac{[R]_2 - [R]_1}{t_2 - t_1}$$

$$R_1 = 0.03 \text{ M}$$

$$R_2 = 0.02 \text{ M}$$

$$t_2 - t_1 = 25 \text{ min}$$

Putting the values, we get:

$$\text{Average rate} = - \frac{0.02 - 0.03}{25}$$

$$\text{Average rate} = - \frac{-0.01}{25} = 4 \times 10^{-4} \text{ M min}^{-1}$$

So, the average rate in minutes will be  $4 \times 10^{-4} \text{ M min}^{-1}$

Now, to find the average rate in seconds we have to divide the above answer by 60.

So, the answer will be:

$$\text{Average rate} = - \frac{4 \times 10^{-4}}{60} = 6.66 \times 10^{-6} \text{ M s}^{-1}$$

Therefore, the average rate in seconds will be  $6.66 \times 10^{-6} \text{ M s}^{-1}$

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**Lingsheng Yao**



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as a main reference throughout the course and kept as a sourcebook after the course had been completed On the other hand this book could not be one of those All you want to know about volumes simply because adhesion is an interdisciplinary and ever growing field For the same reason it would be very difficult for a single individual especially me to undertake the task of writing such a book Thus I relied on the principle that one leaves the truly monumental jobs to experts and I finally succeeded in asking several leading scientists in the field of adhesion to write separate chapters for this collection Some chapters emphasize theoretical concepts and others experimental techniques In the humble beginning we planned to include only twelve chapters However we soon realized that such a plan would leave too much ground uncovered and we resolved to increase the coverage After the book had evolved into thirty chapters we started to feel that perhaps our mission had been accomplished

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