

CHAPTER 4 REVIEW

Chemical Equations and Reactions

Short Review

Write a balanced chemical equation for the reaction described.

1. _____
 - a. balanced chemical equation represents all the following except:
 - i an experimentally established law
 - ii substances by which reactions combine to form products
 - iii identities of reactant and products in a chemical reaction
 - iv relative quantities of reactant and products in a chemical reaction
 - b. According to the law of conservation of mass, the total mass of the
 - i reacting substances is
 - i a always more than the total mass of the products
 - ii always less than total mass of the products
 - iii sometimes more and sometimes less than the total mass of the products
 - iv always equal to total mass of the products
2. Write a balanced chemical equation for the following chemical reactions and label. Put each reaction that will occur. Show the reaction type and complete the chemical equation leaving in the products. Indicate temperature if that is important.
 - a. $\text{H}_2\text{SO}_4 + \text{NaOH} \rightarrow$



Chapter 8 Chemical Reactions Guided Reading Answers

G Psacharopoulos



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