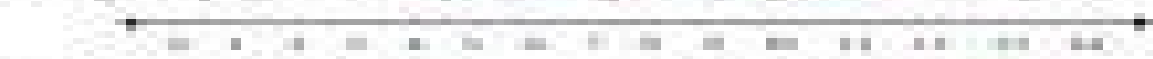


1. Zeichnen Sie die folgenden Komplexe, geben Sie die Oxidationszahl des Zentralatoms an!

- a) $[PtCl_4]^{2-}$ → Oxidationszahl Zentralatom: +2
- b) $[PtCl_6]^{2-}$ → Oxidationszahl Zentralatom: +4
- c) $[PtCl_2(NH_3)_2]$ → Oxidationszahl Zentralatom: +2
- d) $[PtCl_2(NH_3)_4]^{2+}$ → Oxidationszahl Zentralatom: +4
- e) $[PtCl_2(NH_3)_2]$ → Oxidationszahl Zentralatom: +2
- f) $[PtCl_2(NH_3)_4]^{2+}$ → Oxidationszahl Zentralatom: +4
- g) $[PtCl_2(NH_3)_2]$ → Oxidationszahl Zentralatom: +2
- h) $[PtCl_2(NH_3)_4]^{2+}$ → Oxidationszahl Zentralatom: +4



Zeichnen Sie die folgenden Kristallgitter mit Hilfe der Kugelmuster, geben Sie die Gitterkonstanten an!



- a) $NaCl$ → $a_0 = 2 \cdot r_{Cl^-} + 2 \cdot r_{Na^+}$
- b) CaF_2 → $a_0 = 2 \cdot r_{Ca^{2+}} + 2 \cdot r_{F^-}$
- c) Al_2O_3 → $a_0 = 2 \cdot r_{Al^{3+}} + 2 \cdot r_{O^{2-}}$

- d) $NaCl$ → $a_0 = 2 \cdot r_{Cl^-} + 2 \cdot r_{Na^+}$
- e) CaF_2 → $a_0 = 2 \cdot r_{Ca^{2+}} + 2 \cdot r_{F^-}$
- f) Al_2O_3 → $a_0 = 2 \cdot r_{Al^{3+}} + 2 \cdot r_{O^{2-}}$

Chapter 8 Solutions Acids Bases Assessment Answer Key

Rosina Ehmann



Chapter 8 Solutions Acids Bases Assessment Answer Key:

Chapter 8 Solutions Acids Bases Assessment Answer Key Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Chapter 8 Solutions Acids Bases Assessment Answer Key**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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