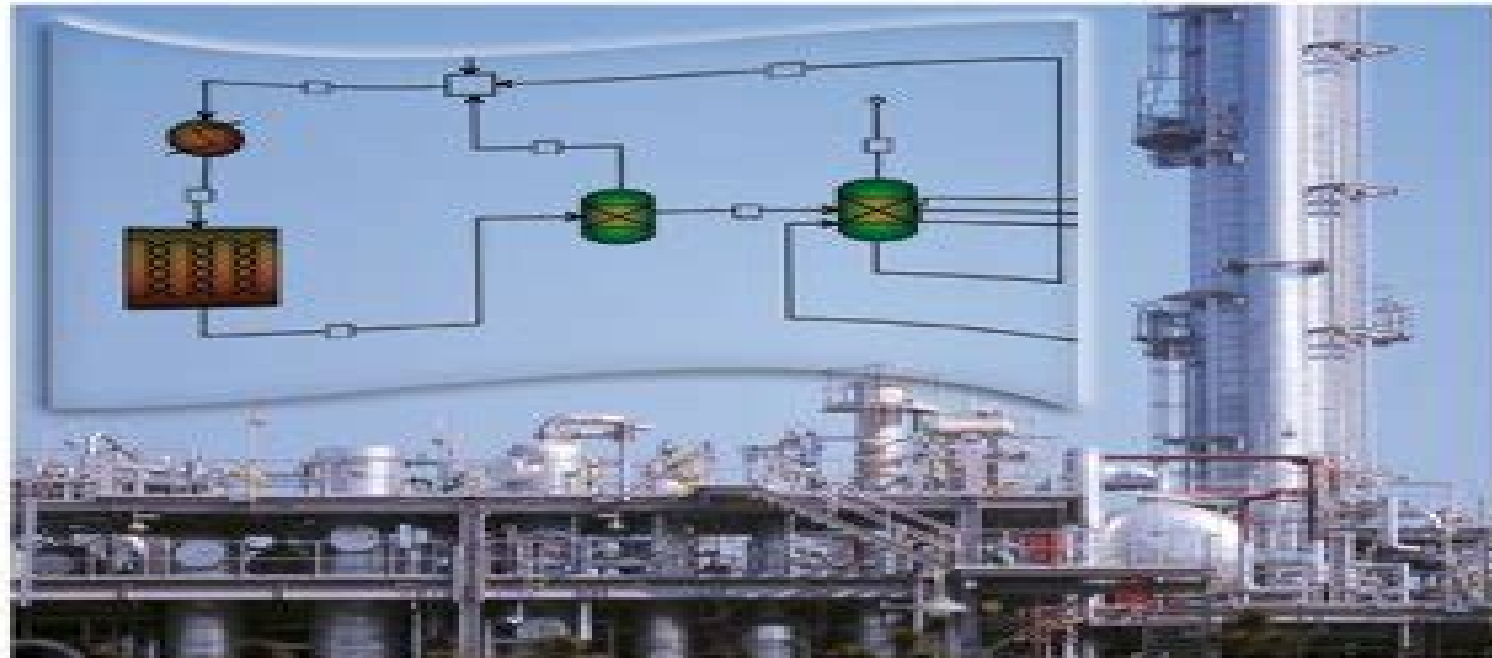


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web the reaction yielded 3 grams of cyclohexene according to the calculations found in appendix a this is a yield of 52 there are a few potential reasons for the yield being lower than ideal there were likely small amounts of substance left behind in previous containers during each transfer

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web jan 1 2021 in fig 1 there are two reaction pathways for the formation of aa one pathway is through the hydrolysis route ka oil is firstly oxidized into cyclohexanone and then cyclohexanone reacts with nitric acid to form 6 hydroxyimino 6 nitrohexanoic acid na which is subsequently hydrolyzed to aa in acid aqueous

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