

Chapter Test A

Chapter: Chemical Equations and Reactions

Do the work provided below the letter of the item or phrase that best completes each statement or best answers each question.

1. The mass number of an atom is the sum of the number of _____ and the number of _____ in the nucleus of the atom.
 - a. electrons; neutrons
 - b. protons; neutrons
 - c. protons; electrons
 - d. neutrons; electrons
2. In the reaction described by a chemical equation, _____ is _____ and _____ is _____.
 - a. oxidized; hydrogenation; a product
 - b. oxidized; iron; product
 - c. reduced; iron; reactant
 - d. reduced; iron; product
3. Which of the following relationships is always true for a chemical reaction?
 - a. The products of a reaction are more stable than the reactants.
 - b. All of the reactants are used.
 - c. The products of a reaction are more stable than the reactants.
 - d. The reactants of a reaction are more stable than the products.
4. There is a negative enthalpy change for a reaction if the products _____ the reactants.
 - a. have a lower enthalpy than
 - b. are at the same enthalpy as
 - c. have a higher enthalpy than
 - d. are at the same enthalpy as
5. In a reaction, a chemical equation _____ the _____ of the reaction.
 - a. shows the enthalpy
 - b. shows the number of products
 - c. shows the number of reactants
 - d. shows the coefficients

Chapter 9 Modern Chemistry Test

KJ Lindholm-Leary



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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Chapter 9 Modern Chemistry Test**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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