

SSC CHSL INSTRUCTIONS FOR TYPING (FOR LDC/JSA AND POSTAL ASSISTANT AND SORTING ASSISTANT)

1. The medium of Typing Test will be Hindi and English. The candidates will have to opt for the medium of Typing Test (ie either Hindi or English) in the Online Application Form.
2. The choice of Typing Test given by the candidate in the online Application Form shall be treated as final and no change in the medium of Typing Test will be entertained later.
3. Candidates opting for English medium should have typing speed of 35 words per minute (wpm) and those opting for Hindi medium should have typing speed of 30 words per minute (wpm). 35 wpm and 30 wpm correspond to about 10500 key depressions per hour and about 9000 key depressions per hour respectively.
4. The speed will be judged on the accuracy of typing on the Computer of a given text passage in 10 minutes
5. TAB Key should be used as given in the passage.

Ssc Typing Test Paper

M Walker



Ssc Typing Test Paper:

The Enigmatic Realm of **Ssc Typing Test Paper**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing in short supply of extraordinary. Within the captivating pages of **Ssc Typing Test Paper** a literary masterpiece penned by a renowned author, readers set about a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting impact on the hearts and minds of those that partake in its reading experience.

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web during this reaction the h^+ ions from the acid and the oh^- ions from the alkali react together to form water as this happens energy is released which is detected by the thermometer

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web 1 sodium hydroxide solution is poured into a beaker of hydrochloric acid which contains a thermometer showing room temperature 2 the beaker now contains sodium chloride and water and the

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web reacting two solutions eg acid and alkali place the polystyrene cup inside the glass beaker to make it more stable

measure an appropriate volume of each liquid eg 25 cm³ place one of the

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web for reactions involving acetic acid or ammonia the measured enthalpy change of neutralization is a few kJ less exothermic than with strong acids and bases for example one source which gives the enthalpy change of neutralization of sodium hydroxide solution with HCl as 57.9 kJ mol⁻¹

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web sep 20 2014 i will illustrate this with some examples when an acid is neutralised by an alkali the reaction is exothermic

eg 1 HCl(aq) + NaOH(aq) → NaCl(aq) + H₂O(l) for which ΔH = 57 kJ mol⁻¹ eg 2 HNO₃(aq) + KOH(aq) → KNO₃(aq) + H₂O(l) for which ΔH = 57 kJ mol⁻¹

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web the enthalpy change of neutralisation is defined as the enthalpy change when solutions of an acid and alkali react together to produce one mole of water for example HCl(aq) + NaOH(aq) → NaCl(aq) + H₂O(aq) you can see that one mole of water is produced when one mole of HCl or one mole of NaOH reacts