

# PRE -LAB, POST -LAB QUESTIONS AND ANSWERS 2022/2023

## pH and Titration Pre-lab questions

1. Explain what is meant by a 2M NaCl solution.

It means that the solution has 2 moles of sodium chloride dissolved in 1 liter of water.

2. In a 5M glucose solution, how many glucose molecules are in one liter?

There are  $5 \times (6.02 \times 10^{23})$  glucose molecules in one liter of a 5M glucose solution.

3. What pH indicator will be used for titrations?

Phenol Red will be used for titrations.

4. What color is the pH indicator under acidic conditions?

It turns yellow when under acidic conditions.

5. What happens to the  $H^+$  concentration when an acid is added to a solution?  
What happens to the pH?

The  $H^+$  concentration increases when an acid is added to a solution. The pH level decreases.

6. If you add phenol red to a solution and it turns pink, what is the pH range of that solution? What titrant (acid or base) would you use to titrate this solution?

The solution is base. so ignore to titrate the solution which means to neutralize it we add an acid to it.

# Titration Pre Lab Answers

**JG Myers**



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