


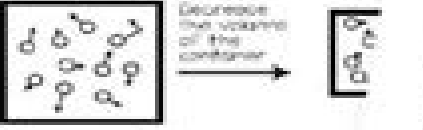
GAS LAWS

Name: _____


Gas molecules are constantly in motion and hit other gas molecules and the walls of the container. For each of the three questions below, write whether the number of collisions per minute increases or decreases and then explain your answer.

1.  Double the number of molecules.

Collisions per minute will _____ (increase/decrease)

2.  Decrease the volume of the container.

Collisions per minute will _____ (increase/decrease)

3.  Increase the temperature.

Collisions per minute will _____ (increase)

Name: _____

Answer the questions below. Use the space under each set of questions for your work.

Class: _____ Date: _____

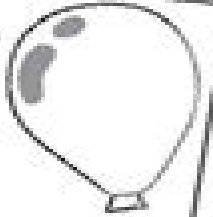
A fun-loving chemist has a balloon at standard temperature and pressure and the balloon has a volume of 3.00 L.



1. What is the initial volume of the balloon?
2. What is the existing temperature?

If the pressure is held constant and the balloon is heated to 400°C, what will the new volume of the balloon be?

3. What is the new temperature of the system?
4. What is the new volume of the balloon?



From the fun-loving chemist - placed it in a box of 4°C. If the...

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