

Chapter 6 Covalent Bonds

Section 6.2 Covalent Bonding**(pages 166–188)**

This section discusses the formation of covalent bonds and the factors that determine whether a molecule is polar or nonpolar. It also discusses attractions between molecules.

Reading Strategy (page 166)

Relating Text and Visuals As you read the section, look closely at Figure 9. Complete the table by describing each type of model shown. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Molecular Models	
Model	Description
Electron-dot diagrams	Dots represent valence electrons.
Structural formula	
Space-filling	
Electron cloud	

Covalent Bonds (pages 166–187)

- Define a covalent bond. _____
- A molecule is a _____ group of atoms that are joined together by one or more covalent bonds. Circle the correct answer: negative neutral positive
- Is the following sentence true or false? In a covalent bond, the atoms are held together by the attractions between the shared electrons and the protons in each nucleus. _____
- Circle the correct answer. Nitrogen has five valence electrons. How many pairs of electrons must two nitrogen atoms share in order for each atom to have eight valence electrons?
 - one
 - two
 - three

Chapter 6 Chemical Bonding Section 2 Covalent Answer Key

Adela S. Baer



Chapter 6 Chemical Bonding Section 2 Covalent Answer Key:

Chapter 6 Chemical Bonding Section 2 Covalent Answer Key Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Chapter 6 Chemical Bonding Section 2 Covalent Answer Key**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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