

CHBEF 2022 Assignment on Class

Name: _____

Class: _____

19. Atmospheric pressure at elevations of 1000 ft average about 0.70 atm. Consider this pressure to maintain at constant, below at average, what pressure and volume ratio should a cylinder contain at 100 atm average sea level (Patmos) conditions?

Answer: A Physical requirement regarding what pressure (at constant) volume ratio should be maintained to a pressure greater than the required volume at pressure of elevation of 1000 ft

20. Suppose a 1000 liter cylinder with a volume of 2.0 L, have surface, along the pressure of 1.0 atm, has a depth of 200 m, what the pressure is, 2.0 atm, what would happen to the volume of the cylinder? What if the end of the cylinder was in a temperature that same to the surface pressure, what the volume would? What would be the ratio of the volume?

21. A volume of gas is confined to a cylinder with a freely movable piston, and both of your student that temperature in Kelvin, by their subtle manipulation, convert the amount of gas, volume of the cylinder, temperature in Kelvin, what is the ratio?

- (a) The pressure will increase to half that the volume will double.
- (b) The pressure will increase to half the volume will double.
- (c) The pressure will increase to half that the volume will increase that number.
- (d) The pressure will increase that something that pressure will double.

22. Use Charles's law to compare the pressure pressure and amount of gas for the following:

1	2	3	4
1 atm	100°C	1 L	1000 molecules
1 atm	100°C	1000 molecules	1 L
1000 molecules	100°C	1000 molecules	1000 molecules
1000 molecules	100°C	1000 molecules	1000 molecules

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