

# Chemical Reactions and Equations

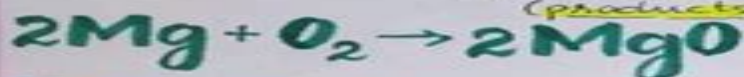
## Chemical Reaction

- change in state
- colour
- gas
- temperature

ex: Magnesium + Oxygen  $\rightarrow$

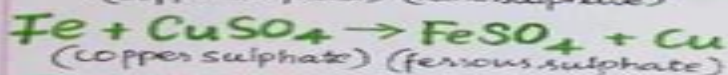
(Reactants)

magnesium oxide  
(products)



## DISPLACEMENT

Reaction

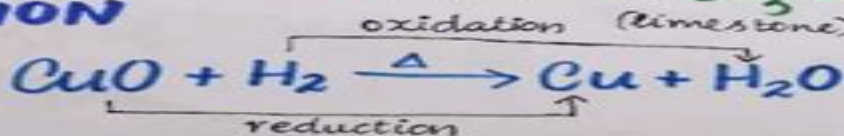


## DOUBLE DISPLACEMENT

Reaction



## OXIDATION & REDUCTION



## Why balance Chemical equation

$\Rightarrow$  Mass can neither be created nor destroyed in a chemical reaction.  
 $\triangleright$  Total mass of elements in the product of a chemical reaction has to be equal to the total mass of the elements in the reactants side.

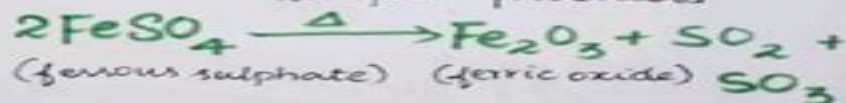
## COMBINATION

Reaction when two or more substances combine to form a single product.



## DECOMPOSITION

Reaction when a single reactant breaks down to simpler products



# Chemistry Notes Chapter 8 Chemical Reactions

**Eric Brian Smith**



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## **Table of Contents Chemistry Notes Chapter 8 Chemical Reactions**

1. Understanding the eBook Chemistry Notes Chapter 8 Chemical Reactions
  - The Rise of Digital Reading Chemistry Notes Chapter 8 Chemical Reactions
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Notes Chapter 8 Chemical Reactions
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry Notes Chapter 8 Chemical Reactions
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Notes Chapter 8 Chemical Reactions

- Personalized Recommendations
  - Chemistry Notes Chapter 8 Chemical Reactions User Reviews and Ratings
  - Chemistry Notes Chapter 8 Chemical Reactions and Bestseller Lists
5. Accessing Chemistry Notes Chapter 8 Chemical Reactions Free and Paid eBooks
    - Chemistry Notes Chapter 8 Chemical Reactions Public Domain eBooks
    - Chemistry Notes Chapter 8 Chemical Reactions eBook Subscription Services
    - Chemistry Notes Chapter 8 Chemical Reactions Budget-Friendly Options
  6. Navigating Chemistry Notes Chapter 8 Chemical Reactions eBook Formats
    - ePub, PDF, MOBI, and More
    - Chemistry Notes Chapter 8 Chemical Reactions Compatibility with Devices
    - Chemistry Notes Chapter 8 Chemical Reactions Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Chemistry Notes Chapter 8 Chemical Reactions
    - Highlighting and Note-Taking Chemistry Notes Chapter 8 Chemical Reactions
    - Interactive Elements Chemistry Notes Chapter 8 Chemical Reactions
  8. Staying Engaged with Chemistry Notes Chapter 8 Chemical Reactions
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Chemistry Notes Chapter 8 Chemical Reactions
  9. Balancing eBooks and Physical Books Chemistry Notes Chapter 8 Chemical Reactions
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Chemistry Notes Chapter 8 Chemical Reactions
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Chemistry Notes Chapter 8 Chemical Reactions
    - Setting Reading Goals Chemistry Notes Chapter 8 Chemical Reactions
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Chemistry Notes Chapter 8 Chemical Reactions

- Fact-Checking eBook Content of Chemistry Notes Chapter 8 Chemical Reactions
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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