

Chemical Analysis – What can you Remember?

1. What colour precipitate do chloride ions produce?
 - a) White
 - b) Cream
 - c) Yellow
 - d) Green
2. What colour do sodium ions go in a flame test?
 - a) Red
 - b) Yellow
 - c) Brick-red
 - d) Lilac
3. Which two solutions are added to test for the presence of halide ions?
 - a) Silver chloride and nitric acid
 - b) Barium chloride and hydrochloric acid
 - c) Silver nitrate and nitric acid
 - d) Barium chloride and sulphuric acid
4. What is the test for oxygen gas?
 - a) Lit splint goes out with a squeaky pop
 - b) Damp blue litmus paper goes red
 - c) Damp red litmus paper goes blue
 - d) Glowing splint relights
5. What colour precipitate to iodide ions form?
 - a) White
 - b) Cream
 - c) Yellow
 - d) Green
6. Which ion produces a brick-red colour in a flame test?
 - a) Sodium
 - b) Calcium
 - c) Copper
 - d) Lithium
7. What is the test for carbon dioxide gas?
 - a) Limewater turns cloudy
 - b) Damp blue litmus turns red
 - c) A glowing splint relights
 - d) A lit splint goes out with a squeaky pop
8. When sodium hydroxide is added to a solution containing iron (III) ions, what colour precipitate forms?
 - a) White
 - b) Blue
 - c) Green
 - d) Red/brown
9. What is the test for chlorine gas?
 - a) Damp blue litmus paper turns red and is then bleached white
 - b) Red litmus turns blue
 - c) A glowing splint relights
 - d) Limewater will turn cloudy
10. Copper (II) ions form what colour precipitate test?
 - a) Blue
 - b) Green
 - c) Red/brown
 - d) Cream
11. Which two solutions are added to test for the presence of sulphate ions?
 - a) Silver chloride and nitric acid
 - b) Barium chloride and sulphuric acid
 - c) Silver nitrate and nitric acid
 - d) Barium chloride and hydrochloric acid
12. What colour do potassium ions go in a flame test?
 - a) Lilac
 - b) Red
 - c) Yellow
 - d) Brick-red
13. What is the test for hydrogen gas?
 - a) Lit splint goes out with a squeaky pop
 - b) Glowing splint relights
 - c) Damp blue litmus paper turns red and then is bleached white
 - d) Limewater goes cloudy
14. Which ion produces a red colour in a flame test?
 - a) Sodium
 - b) Calcium
 - c) Copper
 - d) Lithium
15. What colour precipitate to bromide ions form?
 - a) White
 - b) Cream
 - c) Yellow
 - d) Green
16. What needs to be added to test for the presence of carbonate ions?
 - a) Silver nitrate
 - b) Universal indicator
 - c) Litmus paper
 - d) Dilute acid
17. If sodium hydroxide is added to a compound containing ammonium ions, what gas is produced?
 - a) Oxygen
 - b) Ammonia
 - c) Carbon dioxide
 - d) Hydrogen
18. What colour precipitate does barium sulphate form?
 - a) White
 - b) Green
 - c) Yellow
 - d) Cream
19. What is needed to carry out a flame test?
 - a) Splint
 - b) Glass rod
 - c) Spatula
 - d) Nichrome/platinum wire

Your score out of 19=
Percentage =

Chemistry Unit 2 Test Answers

SJ Ball



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1. Understanding the eBook Chemistry Unit 2 Test Answers
 - The Rise of Digital Reading Chemistry Unit 2 Test Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Unit 2 Test Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Unit 2 Test Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Unit 2 Test Answers
 - Personalized Recommendations
 - Chemistry Unit 2 Test Answers User Reviews and Ratings
 - Chemistry Unit 2 Test Answers and Bestseller Lists
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 - Chemistry Unit 2 Test Answers Enhanced eBook Features
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 - Highlighting and Note-Taking Chemistry Unit 2 Test Answers
 - Interactive Elements Chemistry Unit 2 Test Answers
8. Staying Engaged with Chemistry Unit 2 Test Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Unit 2 Test Answers
9. Balancing eBooks and Physical Books Chemistry Unit 2 Test Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Unit 2 Test Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Unit 2 Test Answers
 - Setting Reading Goals Chemistry Unit 2 Test Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Unit 2 Test Answers
 - Fact-Checking eBook Content of Chemistry Unit 2 Test Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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web questions answers why is ezinma so special to okonkwo why does nwoye convert to christianity why does okonkwo hang himself why are the villagers shocked when okonkwo beats his wife during the week of peace why are the villagers excited whenever the locusts come why does nwoye admire ikemefuna