

### 3. Atoms, elements and compounds



Atomic mass = 23

Proton number = 11

Neutron number = 12 (23 - 11)

mass of electron =  $\frac{1}{1836}$

**ISOTOPE**: atoms of the same element with the same proton number but a different neutron number

- Have the same properties due to the same number of electrons on the outer shell

**Radioisotope**

- testing for leaks in pipes
- treating cancer

#### IONIC BONDING

**Bonding**: oppositely charged ions are strongly attracted by electrostatic force + involves electron transfer

metal + non metal

- metals give electrons to non-metals

**Properties**: strong bonds due to attractive forces

- conduct electricity
- high mp/bp
- charged particles
- soluble in water

#### Macromolecular compounds

**SILICON (IV) OXIDE**

**Properties**



#### COVALENT BONDING

**Bonding**: nuclei of atoms and the shared pair of electrons are strongly attracted by electrostatic forces

non metal + non metal

**Properties**: weak bonds = low mp/bp

- insoluble
- don't conduct

#### Macromolecular compounds



... the strong network of covalent bonding ...  
... the atoms are held together ...  
... the atoms are held together ...

... the atoms are held together ...

A large, detailed diagram or flowchart explaining the properties of macromolecular compounds, including their structure and bonding.

# Igcse Complete Chemistry Notes

**Shasha Hu**



## **Igcse Complete Chemistry Notes:**

## Unveiling the Magic of Words: A Report on "**Igcse Complete Chemistry Notes**"

In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Igcse Complete Chemistry Notes**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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