

Skills Worksheet

Concept Review Chapter 5: Section 1

Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

11

11

ion

octet

12

12

anion

cation

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 12 neutrons, and 10 electrons.
3. In most chemical reactions, atoms tend to reach the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

- a. An anion is an ion with a negative charge.
- a. Valence electrons occupy the outermost energy level of an atom.
- b. The chloride ion and the chlorine atom have different electron configurations.
- c. A cation is an ion with a positive charge.
- d. Many stable ions have an electron configuration of a noble gas.

Answer the following questions in the space provided.

- a. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 13, Group 16, and Group 17 elements compare with those of the noble gases?

They are the same. Group 1 & 2 ions have noble gas configuration of previous row. Group 13, 16 & 17 ions have noble gas configuration in their row.

# Physical Science Chapter 16 Section 1 Answers

**M Woodhall**



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## **Table of Contents Physical Science Chapter 16 Section 1 Answers**

1. Understanding the eBook Physical Science Chapter 16 Section 1 Answers
  - The Rise of Digital Reading Physical Science Chapter 16 Section 1 Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Chapter 16 Section 1 Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical Science Chapter 16 Section 1 Answers
  - User-Friendly Interface

4. Exploring eBook Recommendations from Physical Science Chapter 16 Section 1 Answers
  - Personalized Recommendations
  - Physical Science Chapter 16 Section 1 Answers User Reviews and Ratings
  - Physical Science Chapter 16 Section 1 Answers and Bestseller Lists
5. Accessing Physical Science Chapter 16 Section 1 Answers Free and Paid eBooks
  - Physical Science Chapter 16 Section 1 Answers Public Domain eBooks
  - Physical Science Chapter 16 Section 1 Answers eBook Subscription Services
  - Physical Science Chapter 16 Section 1 Answers Budget-Friendly Options
6. Navigating Physical Science Chapter 16 Section 1 Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Physical Science Chapter 16 Section 1 Answers Compatibility with Devices
  - Physical Science Chapter 16 Section 1 Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Physical Science Chapter 16 Section 1 Answers
  - Highlighting and Note-Taking Physical Science Chapter 16 Section 1 Answers
  - Interactive Elements Physical Science Chapter 16 Section 1 Answers
8. Staying Engaged with Physical Science Chapter 16 Section 1 Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Physical Science Chapter 16 Section 1 Answers
9. Balancing eBooks and Physical Books Physical Science Chapter 16 Section 1 Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physical Science Chapter 16 Section 1 Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Chapter 16 Section 1 Answers
  - Setting Reading Goals Physical Science Chapter 16 Section 1 Answers
  - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Physical Science Chapter 16 Section 1 Answers
  - Fact-Checking eBook Content of Physical Science Chapter 16 Section 1 Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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