Name:_____ Class_____ Period_____ Date_____

Biology Chapter 2 Key Vocabulary

Section 2.1

1. **atom**: the smallest basic unit of matter
2. **element**: one particular type of atom that cannot be broken down by chemical means
3. **compound**: a substance made of atoms of different elements bonded together in a certain ratio
4. **ion**: an atom that gained or lost one or more electrons
5. **ionic bond**: chemical bond formed through the electrical force between oppositely charged ions
6. **covalent bond**: chemical bond formed when atoms share a pair of electrons
7. **molecule**: two or more atoms held together by covalent bonds

Section 2.2

8. **hydrogen bond**: an attraction between a slightly positive hydrogen atom and a slightly negative atom, often oxygen or nitrogen
9. **cohesion**: the attraction among molecules of a substance (ie. water molecules stick to water molecules)
10. **adhesion**: the attraction among molecules of different substances (water molecules sticking to the sides of a glass tube)
11. **solution**: a mixture of substances that is the same throughout (homogeneous mixture)
12. **solvent**: the substance that is present in the greater amount and that dissolves another substance
13. **solute**: a substance that dissolves in a solvent
14. **acid**: a compound that donates a proton when it dissolves in water
15. **base**: a compound that accepts a proton when dissolved in water
16. **pH**: measurement of acidity; related to free hydrogen ion concentration in solution

Section 2.3

17. **monomer**: each subunit in the complete molecule
18. **polymer**: a large molecule, or macromolecule, made of many monomers bonded together
19. **carbohydrate**: molecules composed of carbon, hydrogen, and oxygen; includes sugars and starches
20. **lipid**: nonpolar molecule composed of carbon, hydrogen, and oxygen; includes fats and oils
21. **fatty acids**: chains of carbon atoms bonded to hydrogen atoms
22. **protein**: a polymer made of monomers called amino acids
23. **amino acid**: molecule that makes up proteins, composed of carbon, hydrogen, oxygen, nitrogen, and sometimes sulfur
24. **nucleic acid**: polymer of nucleotides; the genetic material of organisms

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