

Chemistry

Released Test Questions

58. The boiling point of liquid nitrogen is 77 kelvins. It is observed that fog forms at the opening of a container of liquid nitrogen. The best explanation for this observation is

- A. water at zero degrees Celsius is colder than liquid nitrogen and freezes.
- B. the nitrogen boils and then cools to form a fog at the opening of the container.
- C. water trapped in the liquid nitrogen escapes and freezes.
- D. the water vapor in the air over the opening of the liquid nitrogen freezes out.

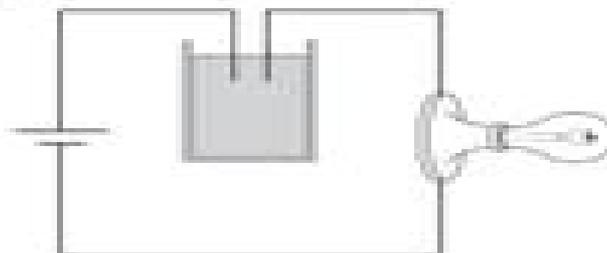
59. The specific heat of copper is about 0.4 joules/gram °C. How much heat is needed to change the temperature of a 30-gram sample of copper from 20.0 °C to 60.0 °C?

- A. 1000 J
- B. 720 J
- C. 480 J
- D. 240 J

60. Equal volumes of 1 molar hydrochloric acid (HCl) and 1 molar sodium hydroxide base (NaOH) are mixed. After mixing, the solution will be

- A. strongly acidic.
- B. weakly acidic.
- C. nearly neutral.
- D. weakly basic.

61.



The above picture shows a light bulb connected to a battery with the circuit interrupted by a solution. When dissolved in the water to form a 1.0 molar solution, all of the following substances will complete a circuit allowing the bulb to light except

- A. hydrochloric acid.
- B. sodium nitrate.
- C. sucrose.
- D. ammonium sulfate.

62. Which of the following is an observable property of many acids?

- A. They become slippery when reacting with water.
- B. They react with metals to release hydrogen gas.
- C. They produce salts when mixed with other acids.
- D. They become more acidic when mixed with a base.

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Robert R. Redfield



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